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Water Monitoring, Assessment & Protection Division

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LANDFILL CLOSURE PLAN

SCANNED

LAURENS CERAMICS SITE LAURENS, SOUTH CAROLINA

This report was reviewed by Div. of Hydro
for consistency w/ closure plans of Bureau of
water. Meno to eng. Sw970 9 68, Efc 10/6/47.

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Plan is consistent of plans rec'd in Div. of Hydro.

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AUGUST 1997

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SECTION 1

INTRODUCTION

1.1 OBJECTIVE OF THE CLOSURE PLAN

The objective of this plan is to present the method for closure of the industrial waste landfill at the Laurens Ceramics site (the former Minnesota Mining and Manufacturing Company/General Electric Company (3M/GE) facility) in Laurens, South Carolina. Preparation of this Closure Plan was done in accordance with the Consent Agreement (No. 96-000-WP) between 3M/GE and the South Carolina Department of Health and Environmental Control (SCDHEC) dated 28 March 1997. The approach to landfill closure in this plan will be protective of human health and the environment, and will minimize the need for further maintenance or mitigation at the landfill.

1.2 SCOPE OF THE CLOSURE PLAN

The scope of this Closure Plan is to describe past operational history at the landfill, summarize existing site conditions, and present an approach to closure that will be protective of human health and the environment. This information and evaluation forms the basis for implementation of the closure method as presented in this plan.

1.3 APPROACH AND ORGANIZATION

The organization of this document reflects the approach developed for closure of the Laurens Ceramics site landfill. Specifically, the sections of this plan and associated approach are as follows:

Section 2 - Review of plant history and the operational history of the landfill.

- Section 3 Presentation of pertinent information gathered during previous investigations and studies.
- Section 4 Presentation of the method of closure.

SECTION 2

BACKGROUND

2.1 FACILITY BACKGROUND

The Laurens Ceramics site is located approximately 2 miles northwest of the town of Laurens, South Carolina on the south side of State Road 14 at the intersection with State Road 24. The plant property encompasses approximately 242 acres and its location is shown on Figure 2-1. Approximately 15 miles to the east is the Little River and approximately 1,000 ft to the south is Reedy Fork Creek.

The facility was constructed during the period 1960-1961 and began operations in 1961 as part of the American Lava Corporation, a subsidiary of 3M. Beginning in 1975, the business was operated as the Technical Ceramics Products Division of 3M. In September 1983, General Electric Ceramics, Inc. (GECI) purchased the facility. GECI sold the facility in March 1988 to AlSiMag, a subsidiary of Eagle Industries. AlSiMag is the current owner and operator of the facility.

From 1961 to 1988, the facility primarily produced textile guides, wear products, and some ceramic electrical substrates used in electrical circuitry, microprocessors, wrist watches, and computers. These ceramics generally contained various amounts of talc, alumina oxides, inorganic inert fillers, and clay. Some of the ceramic formulas also contained low concentrations of barium, including barium titanate and barium carbonate. As part of the normal production process, the ceramic formulations were molded and then ground to proper size and shape. The ceramics were then fired at high temperatures. Roughly 15% of the fired ceramics were ground further in a wet tumbler to a polished state.

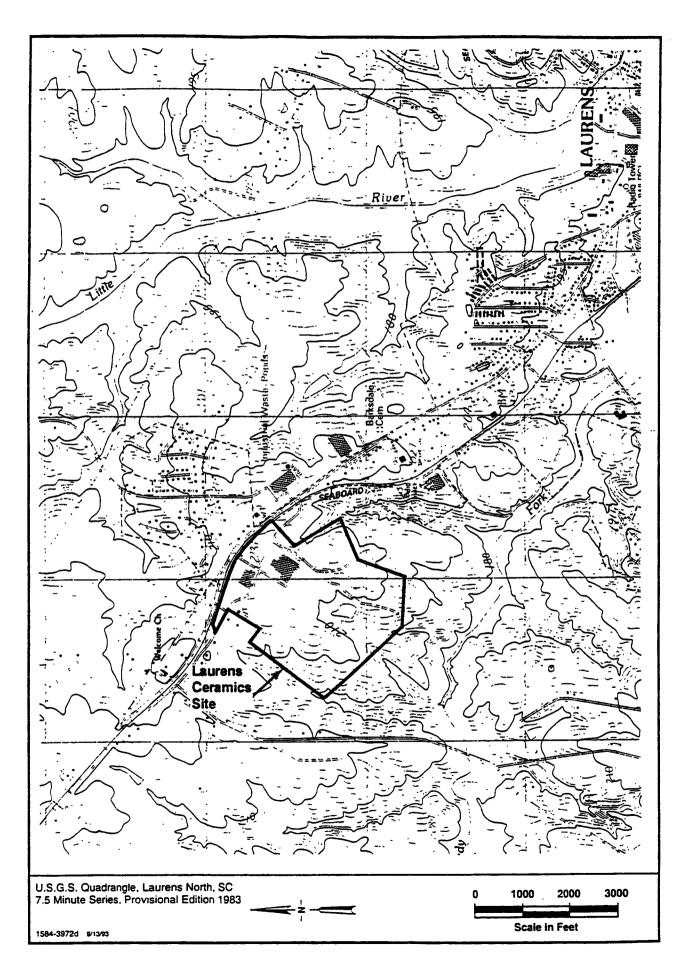


FIGURE 2-1 SITE LOCATION MAP 2-2

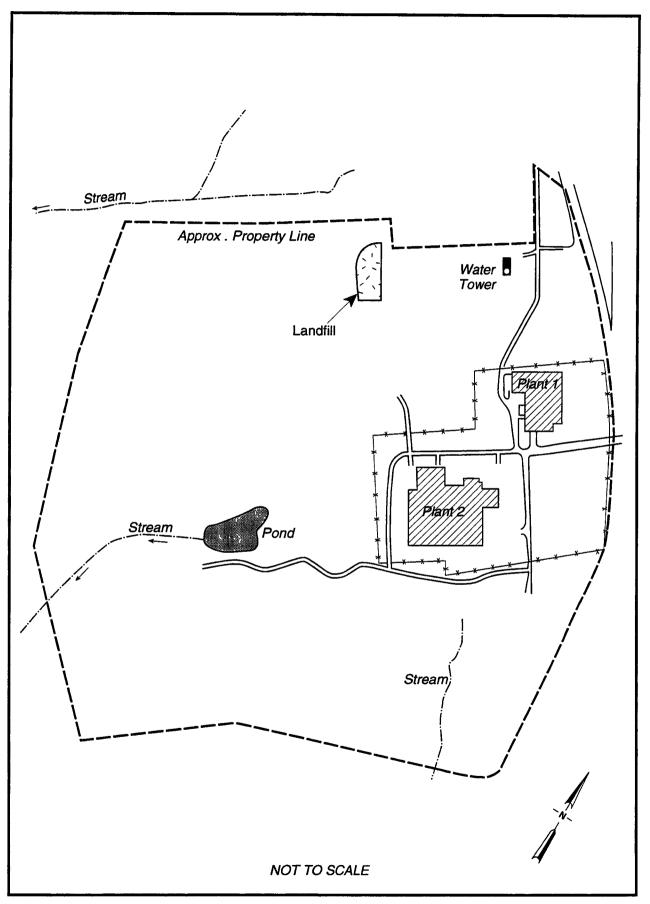
The manufacturing processes that occurred at the facility included body preparation, extrusion mixing, dry pressing, tape casting, glazing, grinding, and post-fire tumbling. Body preparation consisted of processing raw materials into a powder for use in forming products. Extrusion mixing was a basic forming operation which included mixing the prepared powder with water and binder materials prior to extrusion pressing and firing. Dry pressing included mixing of prepared powder with lubricants and pressing in compacting presses. Tape casting involved milling a prepared powder with solvents and casting on a mylar carrier. The glazing operation in the early years of the facility's operation used lead. In subsequent years, a water-based non-toxic glaze system was implemented. Tumbling was a post-firing process to polish and clean the parts. Solid wastes generated as a result of the processes discussed above included:

- Fire (refractory) bricks.
- Fired ceramic parts (off-spec, or otherwise unusable).
- Unfired ceramics.
- Dust collector residues.
- Wet sludge from grinding operations.
- Settling pond ceramic fines.

The manufacturing processes have occurred in two buildings at the facility. These are known as Plants 1 and 2, as shown on Figure 2-2. Plant 1 is the original ceramics production plant which began operations in the early 1960s. Plant 2 began operations in the mid-1960s.

2.2 HISTORICAL USE OF THE LANDFILL

The on-site solid industrial waste landfill is located near the western property boundary approximately 1300 feet south of Highway 14 (see Figure 2-2). The landfill was operated



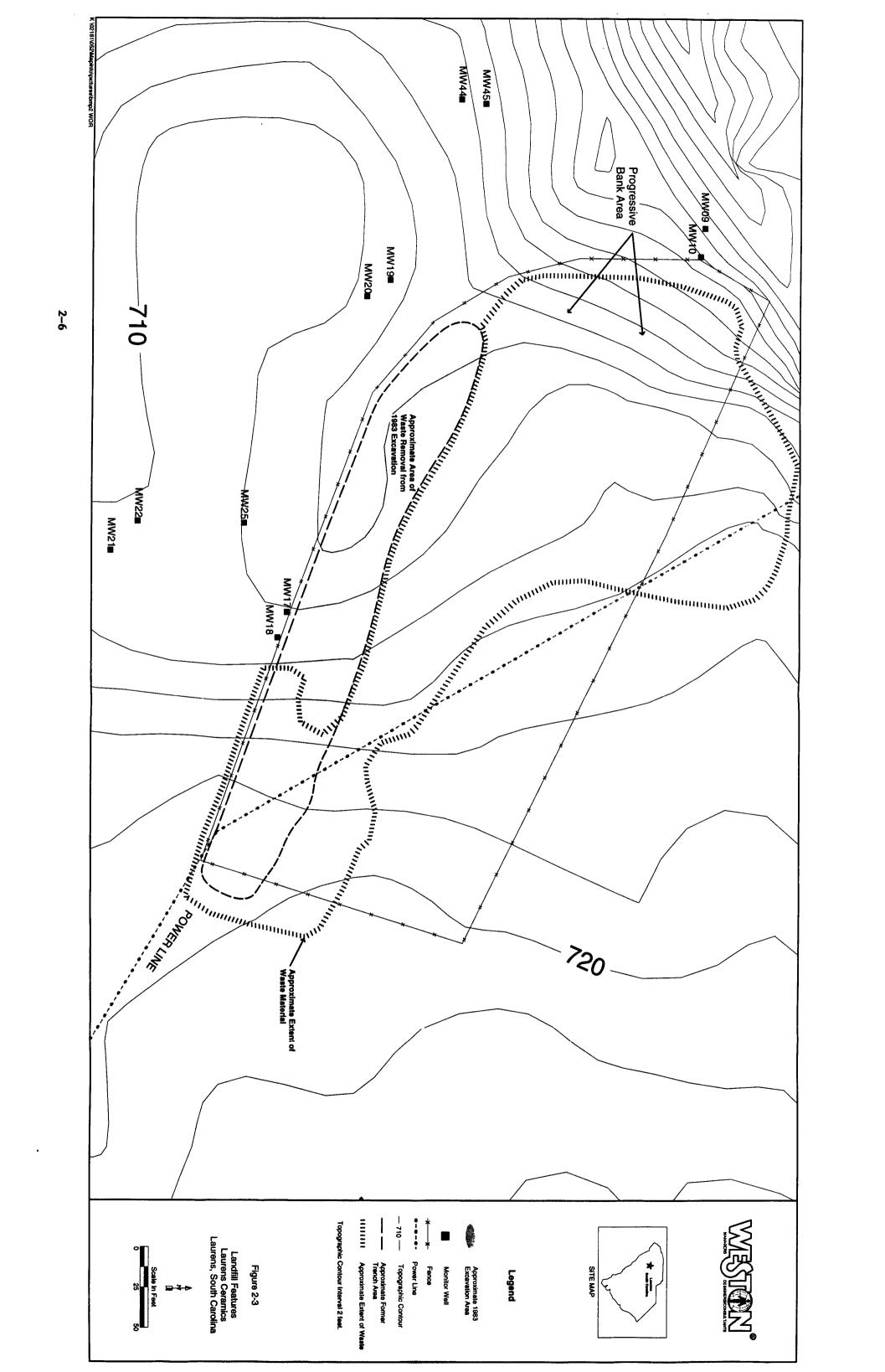
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FIGURE 2-2 PLAN OF THE LAURENS CERAMICS SITE

from 1961 through 1983. Beginning in 1972, the landfill was permitted to accept industrial wastes from the facility. From 1972 through mid-1975, the landfill operated under Permit #57 issued by the South Carolina Pollution Control Authority (SCPCA). From mid-1975 until disposal ceased, the landfill operated under Permit #IWP-123 issued by SCDHEC.

As depicted on Figure 2-3, the landfill area is approximately 2 acres, with wastes encountered up to a maximum depth of approximately 10 feet below ground surface (bgs) and a waste thickness ranging from several inches to 9 feet. The location and distribution of wastes within the landfill is variable since one portion of the landfill was operated as a progressive bank landfill while in another portion wastes were deposited into a trench. Review of the chronology depicted by aerial photography suggests that waste disposal was occurring in 1964 in the progressive bank (northwest portion of the landfill), and appeared to cease in this area before 1980. In the progressive bank portion, waste materials were sequentially deposited over the northwest bank such that the landfill footprint was extended in that direction. In the trench portion, a trench approximately 9 feet in depth was filled. In the 1981 aerial photo, waste disposal in the southeast portion of the trench was distinctly visible. Waste disposal in the trench continued until 1983. Based on observations made during the field investigations, it is estimated that approximately 13,000 cubic yards of waste material were deposited in the landfill during the period it operated.

Generally, 1 to 2 feet of clayey soil with sparse vegetation is currently covering the landfill, although there is an exposed portion on the western boundary (progressive bank portion). Drainage of surface run-off is not well defined and generally follows topography which slopes to the northwest. A small drainage ditch leads from the northwest side of the landfill area and ultimately drains to an unnamed tributary of Reedy Fork Creek. The landfill area is surrounded by a 6 foot chain link fence topped with barbed wire.



The chemical and physical makeup of the material disposed in the landfill has changed over time due to changes in product formulations, changes in procedures, and changes in environmental regulations. From 1972 through early 1981 the following materials were disposed in the landfill under the permit in effect at the time:

- Fired and unfired ceramics.
- Fire brick from furnaces.
- Construction debris.
- Dust collection fines.
- Settling pond fines.
- Paper/wood/cardboard.

Based on information from plant documents and interviews with plant personnel, there was no solvent disposal in the landfill. Some investigations in the vicinity of the landfill have focused on solvents, specifically PCE and TCE. Analysis of the samples collected during test pitting in 1992 showed PCE concentrations less than 0.2 mg/kg in 9 of 10 samples collected. The highest PCE concentration detected (19 mg/kg) had a leachable concentration of only 0.0032 mg/L, several orders of magnitude less than the regulatory criteria (0.7 mg/L). During the 1994 excavation of 10 test pits, VOCs were not detected in the headspace analysis of waste materials. Composition of waste materials is further discussed in Section 3.

With the promulgation of solid and hazardous wastes regulations, waste disposal procedures at the facility were changed. In March 1981, based on preliminary waste analysis using EP Toxicity testing, plant personnel were instructed to segregate wastes resulting from the manufacture of ceramics in which barium-containing raw materials were used. The barium compounds in these raw materials were carbonates, silicates, zirconates, and titanates. The procedure for management of these segregated materials identified disposal at an off-site hazardous waste landfill. In May 1981, SCDHEC was

notified of the past disposal activities, and the intent to dispose these segregated materials off-site. In June 1981 the U.S. Environmental Protection Agency (USEPA) was also notified of disposal activities at the Laurens facility by filing pursuant to §103(C) of CERCLA. In addition to these materials, dust collector fines were also disposed of in the landfill after such materials had been tested in March of 1981 (EP Tox, 1.8 mg/L barium). In 1982, approximately 500 cubic yards of pond sludge were excavated from Plant 2, Pond 1 and placed into the landfill trench. EP Toxicity testing of the pond sludge showed barium levels of 17 and 25 mg/L, a lead level of 0.08 mg/L and no detectable levels of cadmium. The disposal of these sediments in the on-site landfill was approved by SCDHEC in September 1982.

In late 1983, the facility performed additional testing of the dust collector fines originating from the steatite manufacturing process. Subsequent analysis of the fines from this process using EP Toxicity testing indicated a concentration of 580 mg/L barium in the fines sampled. As a result of the sample results, the facility excavated approximately 220 tons of waste from the northwest portion of the trench in the landfill where disposal had been taking place. DHEC was notified of this removal. Samples of the excavated waste materials were tested for EP Toxicity and showed leachable barium concentrations ranging from 300 to 600 mg/L. The excavated materials were managed as hazardous waste and transported to the Chemical Waste Management hazardous waste facility in Emelle, Alabama. Based upon a review of the air permit for the steatite manufacturing process(submitted to SCDHEC in February 1982), it is estimated that there was an average dust pick-up rate of 44 tons per year (based on 10 lb/hr and assuming 24 hr/day operation). Based on this estimate it appears that the excavation of 220 tons would have removed the fines which could have been produced from the steatite process since the amount excavated represents approximately two times (2x) the amount of fines which could have been produced from the steatite manufacturing process from 1980 to 1983. The excavation area was backfilled, graded, and seeded. Following the excavation of the materials, further sampling was conducted to determine if the excavation was complete. EP Toxicity analysis of samples from the base of the trench

showed barium levels consistent with background concentration (0.6 mg/L) indicating the excavation was complete, and also that barium had not migrated. Subsequent groundwater sampling results have further confirmed that groundwater has not been impacted by barium (see Subsection 3.3).

Following, the December 1983 excavation activities, waste disposal activities in the landfill were ceased.

SECTION 3

SITE CONDITIONS

Since 1988, numerous environmental assessments have been conducted to characterize conditions at the Site. These investigations addressed components across the entire plant site, which included the landfill. The following section provides a brief chronology of activities, and a summary of the data collected at the landfill. The data summarized provides physical and chemical characterization of waste materials in the landfill, and characterization of groundwater and surface water quality.

3.1 SITE INVESTIGATION SUMMARY

This subsection presents a brief chronology of the field programs conducted in the landfill area. Table 3-1 outlines sampling activities at the landfill and associated analytical parameters.

3.1.1 Phase I Preliminary Environmental Assessment 1988

Law Environmental, Inc. (LAW) conducted a Phase I Preliminary Environmental Assessment (PEA) investigation in 1988 in support of the property transfer to AlSiMag. The scope of the PEA associated with the landfill included:

- Installation, sampling and water level measurement of 4 groundwater monitor wells.
- A geophysical survey.
- Waste sampling and analysis.

VOC's were detected in groundwater and the initial findings were documented in the *Preliminary Environmental Assessment Report* dated August 1988.

Table 3-1 Chronological Summary of Sampling Activities at the Landfill Laurens Ceramics Site Laurens, South Carolina

Date	Sampling Effort	Analytical Parameters	Description
1988 (LAW)			
1988	Geophysical Survey	l NA	Conducted magnetometer and EM-31 survey
May	Groundwater Sampling	Metals and VOCs	Comprehensive groundwater sampling event
June	Waste Characterization	Metals and VOCs	Analyzed 3 surficial samples of exposed waste materials
June	Groundwater Sampling	Metals and VOCs	Confirmation sampling
September	Residential Supply Well	VOCs, Pesticides, PCBs,	Residential Supply well (Henderson) serving a Trailer Park
•	Sampling	Metals & BNAs	, in the second
December	Groundwater Sampling	Metals and VOCs	Comprehensive Groundwater Sampling
1989 (LAW)		: .: · ::	
June	Waste Characterization	EP Toxicity metals	Analyzed 3 surficial samples of exposed waste materials (in same location as June 1988 sampling)
August	Soil Gas Sampling	VOCs	Soil gas samples collected and analyzed using PETREX tubes
August	Sediment Sampling	Lead and Barium	Two surface sediment samples were composited from ditch immediately downgradient of landfill.
1989	Groundwater Sampling	VOCs	Complete round (MW1-53) of groundwater sampling and analysis
1989	Groundwater Sampling	Metals	Sampled recently installed wells (MW42-53) and analyzed for Total Barium, Lead and Arsenic
1991 (WESTON)			
September/October	Surface Water Sampling	VOCs	Initial surface water sampling
September	Groundwater sampling	VOCs and total metals	Complete round of groundwater sampling (MW1-54)
1992 (WESTON): ::	i i i i i i i i i i i i i i i i i i i		
February	Landfill Trench Characterization	VOCs, total metals & TCLP	12 test pits were excavated.
February	Surface Water Sampling	VOCs	Quarterly surface water sampling
March/April	Groundwater sampling	Metals	Resampled wells near the landfill for total metals
March/April	Groundwater sampling	VOCs	Sampled recently installed wells (MW55-60 & MW63)
May, August, November	Surface Water Sampling	VOCs	Quarterly surface water sampling
1993 (WESTON)	Surface Water Sampling	1003	Quarterly surface water sampling
February, May, August, October	Surface Water Sampling	VOCs	Quarterly surface water sampling
1994 (WESTON)	Surface water Sampling	T VOCS	Quarterly surface water sampning
January-March	Landfill Characterization	Headspace analysis	The second contract the second contract to th
March	Groundwater sampling	VOCs, Metals, Cations,	10 test pits excavated.
March	Groundwater sampling	Anions, & Nutrients	Comprehensive groundwater sampling event
March, June, September, December	Surface Water Sampling	VOCs	Quarterly surface water sampling
1995 (WESTON)		i, water and the contract of t	
January	Groundwater sampling	VOCs	Sampled 2 sentinel wells near the landfill
February, May, November	Surface Water Sampling	VOCs	Quarterly surface water sampling
May 1995	Residential supply well sampling	VOCs	Sampled and analyzed groundwater from 6 residential supply wells near the landfill
May 1995	Groundwater sampling	VOCs, Metals, Cations, Anions, Nutrients, Tritium	Comprehensive groundwater sampling event
August, November	Groundwater sampling	VOCs	Sampled 2 sentinel wells near the landfill
1996 (WESTON)	1	1 2 2 2	
March	Groundwater sampling	VOCs	Sampled 2 sentinel wells near the landfill
March . November	Surface Water Sampling	VOCs	Semiannual Surface water sampling

3.1.2 Phase II Assessment 1988-1991

Based on the results of the PEA, a second phase of activities was implemented to further evaluate the significance of PEA results. The Phase II scope associated with the landfill included:

- Installation of 8 groundwater monitor wells.
- Sampling and analysis of waste, sediment, and groundwater.

The results of the landfill waste analysis showed that while barium was present in the waste, it was not leachable. Analysis of sediments in the drainage ditch northwest of the landfill showed a barium concentration above background levels, but an order of magnitude less than the waste material. Groundwater results were similar to those seen during the PEA. The results of these investigations were documented in the *Phase II Assessment Report* dated April 13, 1990.

3.1.3 Supplemental Investigations 1991-Present

Since 1991, numerous additional investigations have been conducted to characterize landfill materials, soil, groundwater and surface water quality at the Site. The first supplemental investigation was conducted in the Fall of 1991 to provide current data on the spatial extent of VOCs and metal constituents in the surface water and groundwater at the facility. The resulting information was presented in the Supplemental Investigation Work Plan dated December 1991.

The Work Plan summarized existing site data, presented the current conceptual site model, provided a scope of work for additional investigative tasks to support site remedial planning, and further site characterization. These additional site characterization activities at the landfill area included groundwater sampling and excavation of test pits in

the landfill to physically and chemically characterize waste materials. These data served as the basis for numerous subsequent investigations.

Activities conducted in the landfill area during the subsequent investigations have been summarized in Table 3-1 and include the following:

- Sampling, and analysis of waste materials and soils from test pits excavated within the landfill.
- Installation of six additional monitor wells at the landfill.
- Multiple groundwater and surface water sampling and analysis programs, and water level monitoring events.
- Sampling and analysis of select residential supply wells within 1/2 mile of the landfill.

3.2 WASTE CHARACTERIZATION

3.2.1 Physical Characterization

During the PEA, several tasks were conducted to physically characterize the waste and define the landfill boundaries including a site reconnaissance, a geophysical investigation consisting of magnetometer and EM-31 conductivity surveys, and sampling of waste materials.

During the PEA site reconnaissance, it was reported that the northwestern toe of the landfill was exposed. An empty crushed drum, bricks, lumber, ceramic powder and other debris were visible. Also, small fired ceramic pieces were observed in the small drainage ditch located near the northwestern toe of the landfill.

The magnetometer survey conducted as part of the PEA detected an area of magnetic anomalies which were interpreted as buried objects in the landfill. An overhead utility

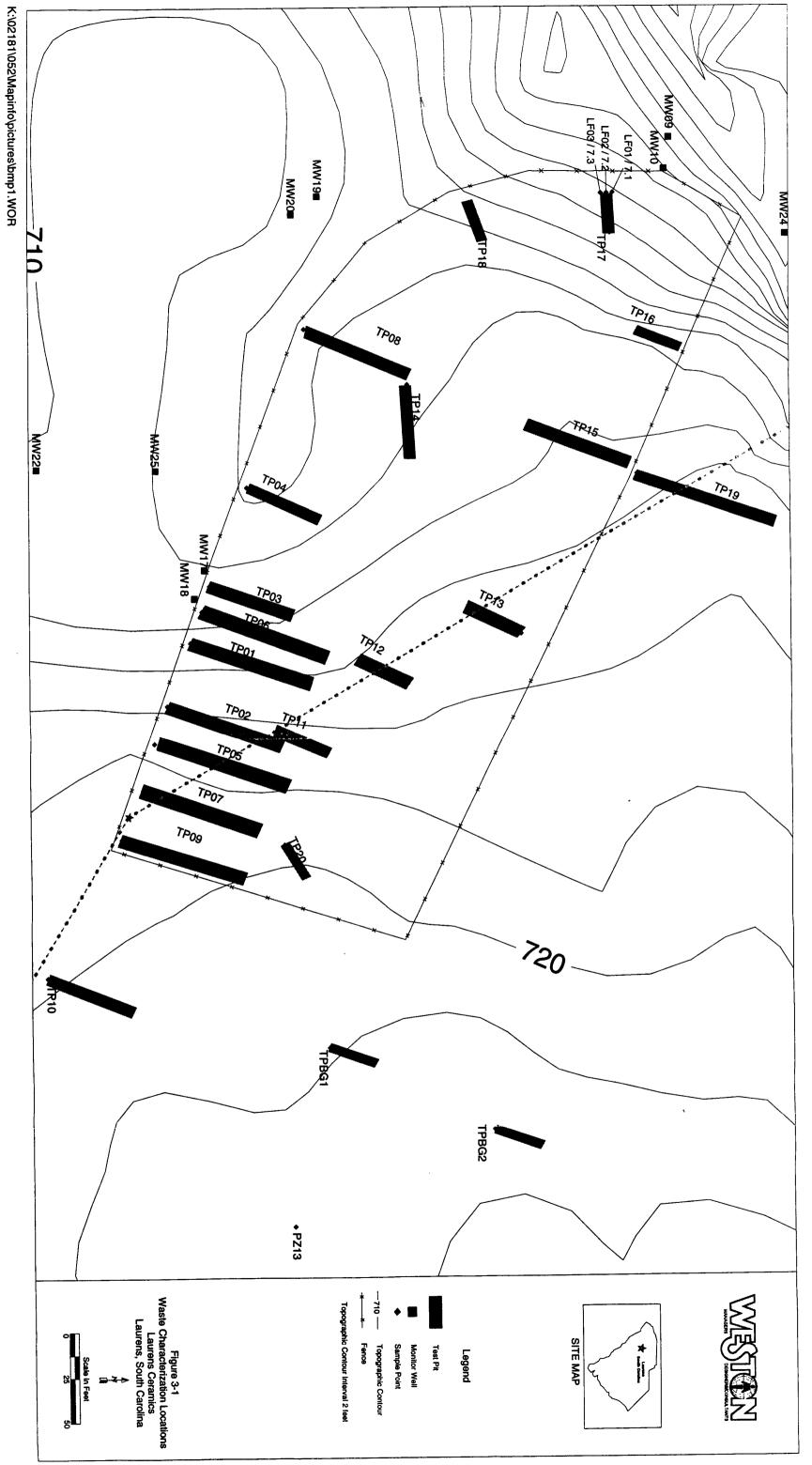
line was noted to have produced electrical interference, obscuring the landfill boundary definition. Data analysis indicated the presence of a large magnetic anomaly trending SW to NE across the landfill which corresponds to the location of the overhead utility line. In addition, several small magnetic anomalies were identified within the landfill. These small anomalies were suggested to potentially represent isolated metal pipes.

During the follow up investigation in February 1992, two background test pits (TPBG1 and TPBG2) and ten test pits (TP1 to TP10) were excavated in the general area of the former disposal trench (located in the southwestern portion of the landfill). These test pits were constructed to confirm the location of the former trench, to investigate the contents and orientation of the trench and nearby magnetic anomalies previously identified in the PEA, and to collect samples of the waste material and underlying soils. Figure 3-1 indicates the location of these test pits. The descriptions of soils and waste materials and cross sections illustrating materials encountered at the test pit locations are presented in Appendix A. Waste materials encountered included:

- A dry powdery and moist waste material (ceramic fines).
- Fired ceramic pieces.
- Miscellaneous construction materials (including bricks, rebar, wood, metal piping, sheet metal, wires, and steel beams).

The results of the test pit excavations conducted in 1992 confirmed the location of the former disposal trench along the southwestern portion of the landfill (along a line extending southeast to northwest). Some limited fine ceramic waste material was encountered in the southeastern portion of the former trench. In the northwestern portion of the former trench, the backfill soils from the 1983 excavation and removal of waste materials were encountered.

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In the southeast portion of the former trench 1-2 feet of clayey soil was generally encountered as cover material at the surface of each test pit. The cover soils of the former trench were underlain by fine ceramic waste material with a thickness ranging from several inches to 9 feet. The waste thickness appeared to reach its deepest point along the central axis of the trench (i.e., the axis running SE to NW). The majority of the test pit excavations also uncovered miscellaneous scrap metal (including rebar, metal piping, steel beams, wires, etc.) at the edge of the former trench. The locations of these scrap metals roughly correspond to areas of geophysical anomalies reported in 1988.

During the Spring of 1994, an additional ten test pits (TP11 to TP20) were excavated in other areas of the landfill exclusive of the trench (i.e., progressive bank and northern portions) to further characterize waste materials and investigate the remaining isolated magnetic anomalies previously reported. Figure 3-1 indicates the location of these test pits.

The materials encountered during the 1994 test pitting were predominantly ceramic brick, cinder block, scrap wood, scrap metal, and soil fill. Several test pits encountered some localized areas of thin dry waste material (ceramic fines). No elevated VOC readings were detected during routine monitoring of the waste material, fill soils, and underlying soils. The summary logs and cross sections of test pits TP11 to TP20 are presented in Appendix A.

Results of the ten test pits excavated in the Spring of 1994 also identified miscellaneous scrap metal in the majority of the excavations. Many of these locations roughly correspond to the areas of geophysical anomalies identified previously, which were not located under an overhead utility line.

3.2.2 Chemical Waste Characterization

In 1988, three surficial samples of exposed waste materials were collected from the northwestern toe of the landfill (part of the progressive bank portion) at sample locations 7.1 to 7.3 (see Figure 3-1) and analyzed for total metals. In 1989, during the Phase II investigation, additional waste material samples at locations LF1 to LF3 were taken at the same locations as 7.1 to 7.3 and analyzed for metals via the EP toxicity test. A composite sediment sample was obtained from a drainage ditch at the base of the landfill and analyzed for total lead and barium. The analytical results for these samples are presented on Table 3-2.

During the 1992 excavation of the test pits in the former trench, 5 samples were submitted for TCLP analysis and 13 samples were analyzed for total metals and VOCs. The analytical results for these samples are presented on Table 3-2.

In summary, the results from the 8 locations where total and leachable barium testing was performed show that while barium was present, it was not leachable above the RCRA regulatory level except in one sample. The testing results also show that while PCE was present, it was present below 0.2 mg/kg in 9 of 10 samples, and it was not leachable above the RCRA standard for identifying a characteristic waste in any sample.

3.3 GROUNDWATER AND SURFACE WATER

Since 1988, 19 monitor wells have been installed in the landfill area; the location of these wells is depicted on Figure 3-2. A review of the historical groundwater quality data, reveals the following:

Summary of Landfill Soil and Waste Sampling¹ Laurens Ceramics Site **TABLE 3-2**

					PCE			Barium	
						Regulatory			Regulatory
				Total	Leachate ²	Limit	Total	Leachate ²	Limit
Location	Sample Type	Depth	Date	(mg/kg)	(mg/L)	(mg/L)	(mg/kg)	(mg/L)	(mg/L)
7.1A	Surficial	NS	1988				9300		
7.18	Surficial	SN	1988				81		
7.10	Surficial	NS	1988				9500		
LF-1	Waste material	NS	1989					SQ.	100
LF-2	Waste material	NS	1989					SD.	100
LF-3	Waste material	NS	1989					ND	100
TP1	Waste material	3.5	1992	19	0.0032	7.0	17300	11.7	100
TP2	Trench base	13	1992	0.098	ΩN	2.0	32.1	0.483	100
TP3	Soil fill	6	1992	0.0026			QN		
TP4	Trench base	11	1992	0.0035			74.2		
TP5	Waste material	3	1992	0.04	<u> </u>	2.0	9440	504	100
TP6	Trench base	8	1992	0.11	QN	2.0	131	1.79	100
TP7	Trench base	5	1992	0.14	QN	2.0	QN	0.535	100
TP8	Soil fill	4	1992	ND			63.8		
ТР9	Trench base	6	1992	0.033			9		
TP10	Trench base	3.5	1992	0.0015			Q		
TPBG	Background	2	1992	ND			QN		
TPBG	Background	3.5	1992	QN			142		
TPBG	Background	9	1992	QN			QN		

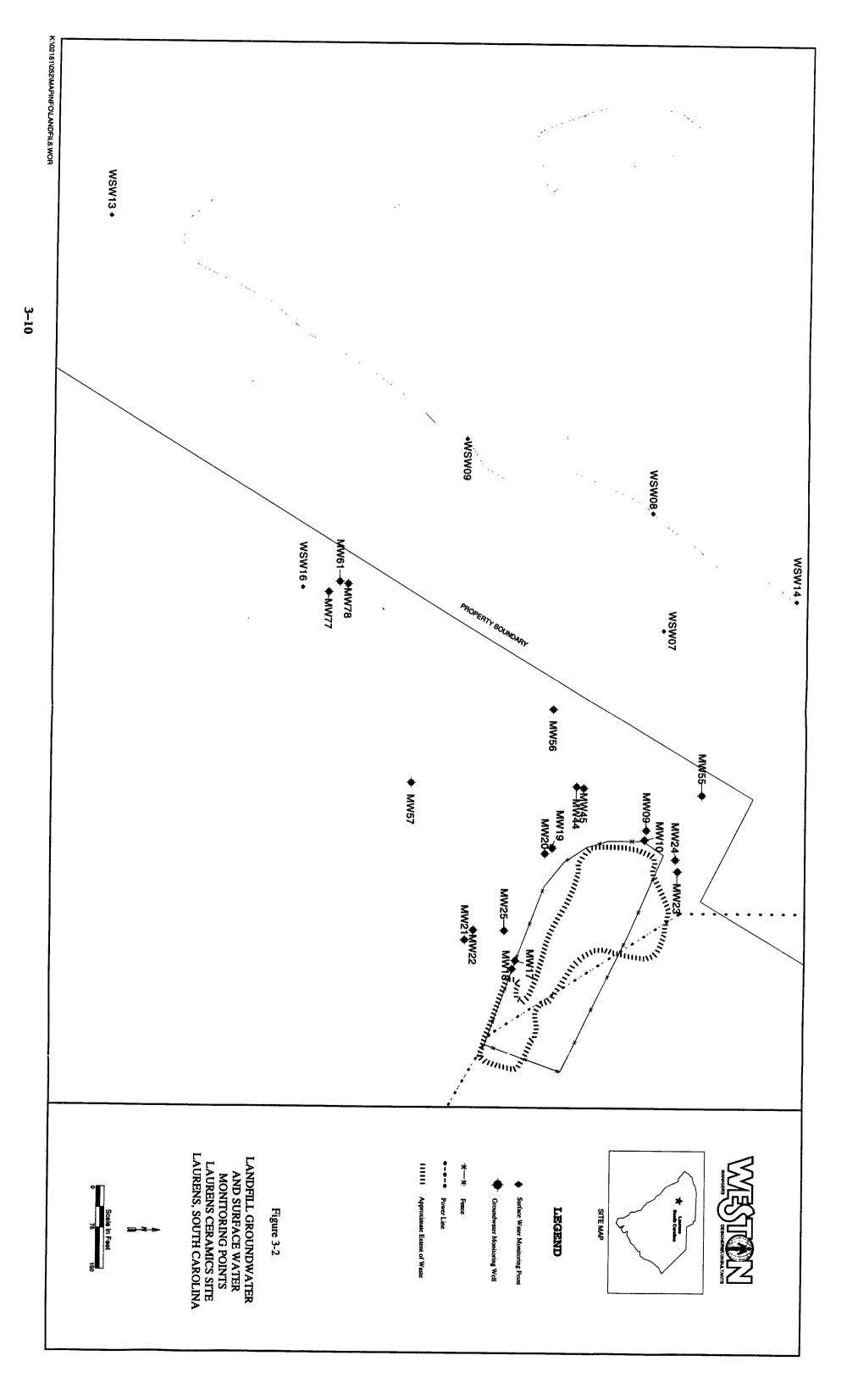
NOTES:

NS - Not specified.

Blanks indicate analysis was not performed.

- 1 A complete summary of all analytes is presented in Appendix B.2 Leachate analysis by TCLP procedure, unless otherwise noted.3 Leachate analysis by EP Toxicity prodecure.

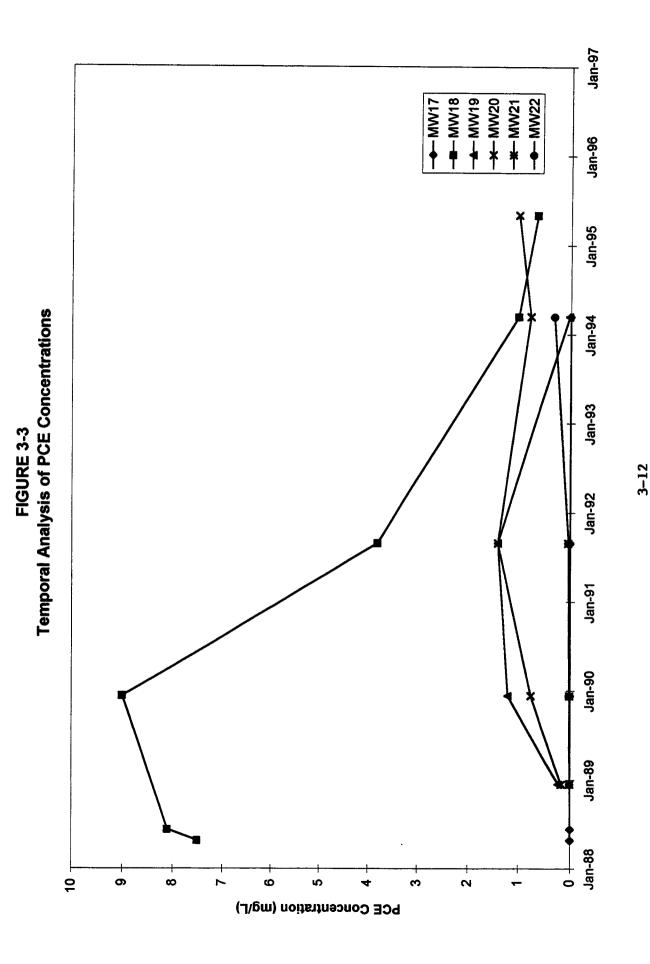
3-9



- Groundwater flows in a southwesterly direction.
- Metals, including barium, are not migrating from the landfill into groundwater. This is evidenced by the latest groundwater quality data.
 Historical groundwater quality data is presented in Appendix B.
- VOCs, including PCE, are present in groundwater upgradient and downgradient of the landfill.
- The data indicates that the landfill is not acting as a source of VOCs. As shown on Figure 3-3, the concentrations of PCE has been decreasing since 1991, especially at MW18 (at the downgradient landfill boundary) where the PCE concentrations have decreased from approximately 9 mg/L in 1990 to approximately 0.6 mg/L in 1995. A similar decrease has also been observed in downgradient well MW19 where PCE has decreased from 1.4 mg/l in 1988 to 0.026 mg/L in 1994. The PCE concentrations at other downgradient wells is remaining relatively constant.

Most of the surface water runoff from the landfill is collected at the drainage ditch that extends from the northwestern toe of the landfill (near MW09 and MW10) to a small (0 to 2 gpm) spring (surface discharge of groundwater) located approximately 300 feet to the west of the landfill (see Figure 3-2). The spring flows into the unnamed tributary of Reedy Fork Creek.

Since 1991, the surface water locations to the west have been sampled on numerous occasions. Quarterly sampling and analysis of surface waters on or immediately adjacent to the Site was initiated in August 1992. In November 1995, semiannual sampling of surface water stations was adopted. Six surface water sampling points (WSW07, WSW08, WSW09, WSW13, WSW14, and WSW16) have been used to monitor the effect of runoff from the landfill and surface discharge of groundwater (see Figure 3-2). In the unnamed tributary, no water quality impacts are observed. The exception to this is WSW16. This monitoring point coincides with the surface discharge of groundwater at a spring and the VOC concentrations observed are similar to groundwater quality in this



area. This is expected because the spring at WSW16 is a surface discharge of groundwater. The complete analytical results from these locations are presented in Appendix C. The analytical data demonstrate that surface water runoff from the landfill is not a concern. The following summarizes the analytical data collected at these surface water locations:

	PCE		Barium (total)	
Location	Frequency of Detection	Range of Detection (ug/L)	Frequency of Detection	Range of Detection (mg/L)
WSW07	2/14	ND-1.9	1/1	0.05
WSW08	1/15	ND-14	NA	NA
WSW09	1/18	ND-1.8	0/1	ND
WSW13	1/15	ND-6.1	0/1	ND
WSW14	0/13	ND	0/1	ND
WSW16	3/3	13-54	NA	NA

NOTES:

ND - Not detected above detection limit.

NA - Not analyzed.

SECTION 4

CLOSURE PLAN

4.1 INTRODUCTION

The on-site landfill has not been used for disposal of waste since 1983. In addition a significant amount of waste was excavated and sent for off-site disposal in 1983. It has been covered with soil of varying thickness and vegetation has emerged over the area. In order to continue to protect human health and the environment, and to provide long-term integrity of closure and improved environmental management, a plan for enhanced final closure of the landfill has been developed.

Based on a review of the site conditions, DHEC Regulatory Guidance for landfill closure, and the closure activities previously approved by the DHEC and implemented for the wastewater ponds at Plants 1 and 2 (WESTON, 1995), WESTON has developed this plan for final closure of the landfill. The key components of the closure activities are as follows:

- Site preparation activities to remove surface and deep-rooted vegetation, and setup work facilities (access roads, etc.).
- Consolidation of materials into a uniform footprint for the landfill.
- Slope regrading.
- Grading to achieve positive drainage contours.
- Stabilization of any remaining sediment materials in the former trench.
- Installation of a cover system that includes a geosynthetic clay liner cap (GCL), and soil cover.
- Restoration activities including surface grading to control surface water runon and run-off, and re-installation of a fence around the perimeter of the landfill.

Revegetation of disturbed areas.

4.2 METHODOLOGY

Based on the results of the site conditions investigation, the landfill is not an on-going source of VOCs or barium to groundwater or surface water. As a result, the preferred approach for long term management of the landfill is in-place closure and improvement to the existing cover. The existing cover will be improved to minimize maintenance needs, provide positive drainage and reduce infiltration. The in-place closure approach for the landfill will closely follow the methodology that was successfully implemented for the on-site wastewater ponds at Plants 1 and 2, and approved by DHEC.

Test trenching performed during the site investigation indicated that most of the former trench was backfilled with soil. However, any localized areas or pockets of sediments/fines in the trench area will be stabilized using the same Portland Cement recipe that was used successfully on the sediments associated with the on-site wastewater pond closure. This stabilization will provide the following:

- Consistency with the methodology used previously for the approved pond sediment closure and;
- Chemical stabilization and immobilization of any residual barium that might be associated with any sediment-type wastes remaining in the trench.

Based on the pond closure experience, it is expected that a stabilization recipe of 15 weight percent Portland Cement Type 1 (PCI) will be used.

Based on the operational history of the landfill (see Section 2), and the results of previous field work, there are locations within the current landfill footprint that have only near-surface, thin layers of fine ceramic waste material. To provide a more uniform final footprint and cover system, these small waste areas will be consolidated into the central

portion of the landfill. Further, relocation of materials from the steep slope area at the northern end of the landfill (progressive bank portion) may be necessary to reduce this slope and improve stability of the final cover system. These materials will be consolidated in the central portion of the landfill to provide the fill needed to improve contours and achieve positive, controlled drainage. It is estimated that 500 to 1,000 yd³ will be consolidated from the progressive bank portion and other areas of the landfill. These consolidation and rough grading activities are depicted on Figure 4-1.

The design of the final cover system will be the same as the one used for the previous closure of the on-site wastewater ponds. This cap system design was previously reviewed and approved by the solid waste/landfill group at DHEC. The following are the key design specifications of the cover system:

- Grading: The landfill regrading will allow for positive drainage away from the landfill cover, provide stable slopes and aesthetically conform, to the extent possible, to the natural terrain.
- Sub-base: Prepared sub-base suitable for placement of the cap.
- Cap: A GCL, with a permeability of approximately 5 x 10⁻⁹ cm/sec, will be placed over the prepared sub-base.
- Soil Cover: 18 inches of clean soil will be placed over the GCL. This soil
 will provide protection for the cap and a physical barrier over the waste to
 prevent direct contact or impact to surface water run-off.
- Topsoil: A 6" thick top layer of soil will be provided which will be suitable for revegetation and establishment of vegetable cover.
- Vegetation: Grasses, such as rye, fescue, bermuda or other low maintenance type grass will be seeded to provide a vegetative cover.



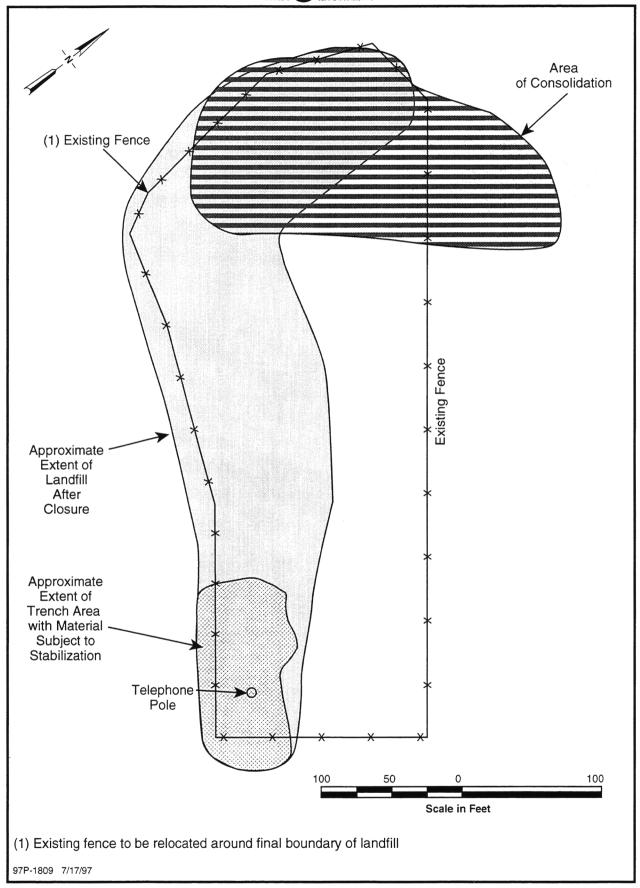


FIGURE 4-1 LANDFILL CLOSURE ACTIVITIES

A cross-section of the cover system is presented in Figure 4-2.

4.3 GROUNDWATER MONITORING

Following completion of the capping and closure activities, groundwater downgradient of the landfill will be monitored annually. Groundwater samples will be collected from the shallow wells MW-18 and MW-20 and analyzed for soluble metals and VOCs. Analytical results will be provided to SCDHEC.

It should be noted that as part of the Consent Agreement, a Landfill Groundwater Assessment is planned. During this assessment, the groundwater monitoring program for the landfill will be reviewed and re-evaluated in the context of the site wide groundwater assessment. A final groundwater monitoring program for the landfill will be proposed at that time.

4.4 **SUMMARY**

This approach for in-place closure of the landfill offers the following advantages:

- The cap design was previously reviewed and approved by DHEC for the pond closure.
- The use of the GCL cap as part of the final cover system will significantly reduce infiltration compared to the current soil cover.
- Consolidation of materials into the central portion of the landfill will improve contours and allow positive drainage, optimize coverage of the cap system, create a smaller, more uniform landfill footprint and improve slope stability.



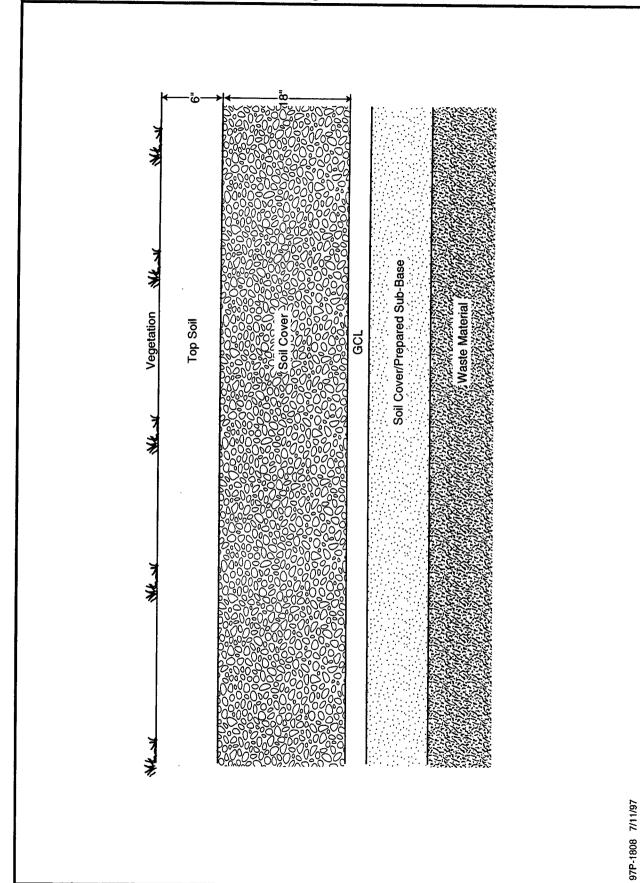


FIGURE 4-2 LANDFILL COVER CROSS SECTION

- In-place stabilization of remaining sediment type wastes in the trench area will
 further immobilize any barium that is present. This adds an additional level of
 protection to the closure even though groundwater, surface water, and waste
 analysis indicate that barium has not been migrating from the waste materials.
- The closure methodology is consistent with the approach employed successfully for closure of the on-site wastewater ponds.

4.5 SCHEDULE

In accordance with the provisions of the Consent Agreement, the closure planning and scheduling activities will begin within 30 days of SCDHEC approval of this plan. Closure activities will be completed within 120 days of commencement of field construction.

4.6 **CLOSURE CERTIFICATION**

Within 60 days of completion of closure of the landfill, a certification of closure, prepared by a registered professional engineer, will be submitted to SCDHEC via registered mail indicating that the landfill has been closed in accordance with this plan. The engineer or designated representative will be present during the activities related to closure of the landfill. The engineer will review the data for the certification of closure. The following information will be maintained as part of closure documentation:

- Approved closure plan.
- Date that closure activities commenced.
- The identification of the primary contact for the closure activities, including name, address and telephone number.
- Documentation of any deviation from the closure plan, including letters or correspondence from SCDHEC approving modifications to the plan.

- Daily inspection or field summary reports documenting closure events.
- Field notes of inspections, sample collection, etc.
- Records of any stabilization activities.
- Records and test results for any samples collected as part of the closure activities.
- Documentation of sampling and test methods.

APPENDIX A LANDFILL TEST PIT DESCRIPTIONS AND CROSS-SECTIONS

GENERALIZED SUMMARY OF TEST PIT RESULTS - FEBRUARY 1992 LAURENS CERAMICS FACILITY LAURENS, SC

Test Pit #	Depth (Ft)	OVA Headspace	Material
TP-1	0-2 2-7	BG 45-100	Fill: sandy clay Fill: Waste material comprised primarily of moist white waste material. Concrete waste and metal piping were detected in the eastern portion of the test pit. Note waste materials were detected 2-7 ft bgs next to fence and in northeastern portion of the test pit. A two foot thisck zone of moist white/gray waste materials were also encountered at approximately 8-10 feet bgs under the loose soil fill. Waste adjacent to fence was sampled at 3.5 feet deep. Sample consisted of moist gray and white ceramic fines.
	7-11	45	Natural Soil: silty sand
TP-2	0-2 2-9	BG 12	Fill: sandy Clay Fill: Waste material: moist white/gray/brown fines. White waste material mixed with concrete asphalt, metal piping and plastic trash bags were encountered in the northeastern portion of the excavation.
	9-13	5-12	Natural Soils: clayey sands.
TP-3	0-7 7-15	BG BG-1	Fill: clayey sand Natural soil: silty sands
TP-4	0-5 5-11	BG 2-6	Fill: clayey fine sand Fill: Clayey sand. Concrete and asphalt debris were encountered in the northeastern portion of the test pit.
TP-5	0-1.5 1.5-4	BG BG-7	Fill: clayey sand Fill: Moist bluegreen and gray waste ,material; including concrete and rebar and fired ceramics which became more abundant in a northeastern direction. Sampled at 3 feet. Sample consisted of moist bluegreen and gray waste material.
	4-7	BG-4	Natural Soil: clayey sandy silt
TP-6	0-7	BG	Fill: clayery silty sand. A 1-2 foot thick zone of dry white waste material and misc. construction materials encountered in the northeastern edge of the test pit.
	7-12	2	Natural Soil: Silty sand
TP-7	0-1.5 1.5-4.5	BG BG-4	Fill: silty clayey sand Fill: Moist bluegreen and gray ceramic fines. Dry white and gray waste material including metal piping, concrete and rebar were encountered in the northeastern edge of the test pit.
	4.5-11	BG-6	Natural Soil: Clayey silt
TP-8	0-6	BG	Fill: clayey sand. In the western edge of the test pit a dry powdery ceramic fines with brick and concrete waste was encountered
	6-8	1	Natural soil. Clayey silt
TP-9	0-3 3-7	BG-2 BG-2	Fill: clayey silty sand Fill: Moist bluegreen and white fine ceramic material. A dry white fine ceramic waste material with concrete and asphalt debris was encountered in the northeastern portion of the test pit.
	7-9	BG-4	Natural Soil: Clayey sand
TP-10	0-1	BG	Fill: clayey sand including small chunks of small moist white and gray ceramic fine material.
	1-4	BG-1	Natural Soil: Clayey sand

GENERALIZED SUMMARY OF TEST PIT RESULTS - FEBRUARY 1992 LAURENS CERAMICS FACILITY LAURENS, SC

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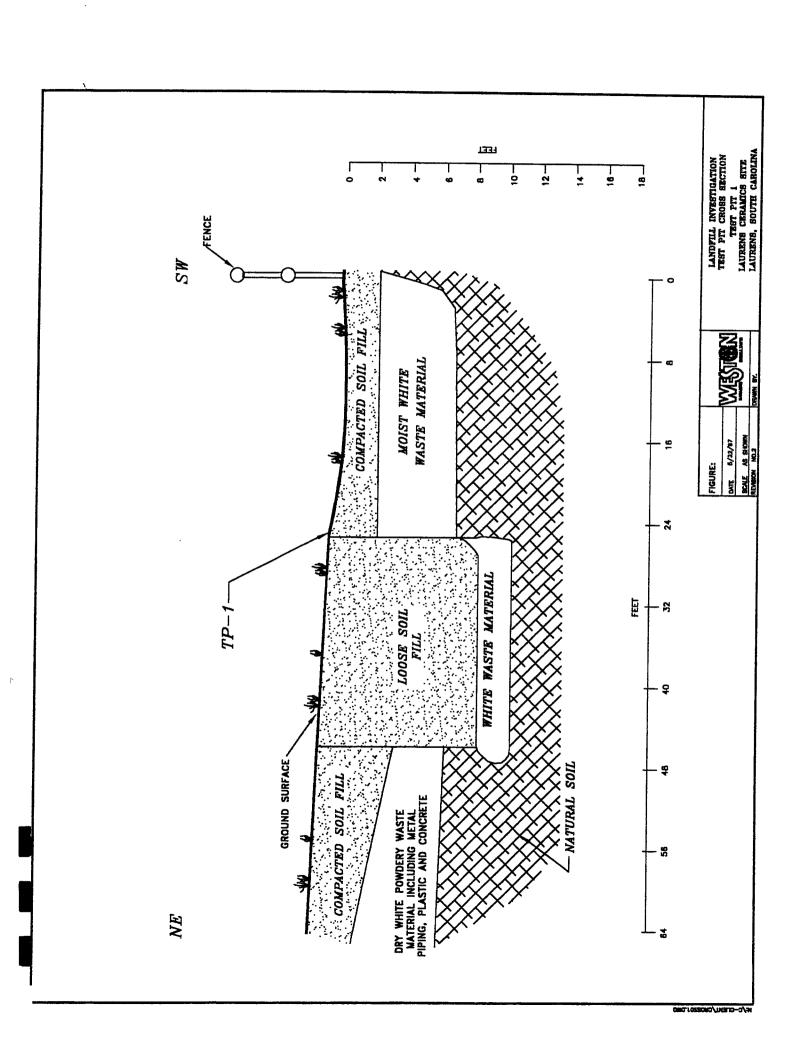
Test Pit #	Depth (Ft)	OVA Headspace	Material
TPBG-1	0-1	BG	Natural Soil: Redbrown clayey silt with a thin layer of organic material at the surface.
	1-3	BG	Natural Soil: Redbrown clayey silt.
	3-4	BG	Natural Soil: Redbrown clayey silt with trace quartz gravel.
	4-6	BG	Natural Soil: Redbrown and white clayey silty sand with trace quartz gravel
TPBG-2	0-1	BG	Natural Soil: Redbrown clayey silt with a thin layer of organic material at the surface.
	1-3	BG	Natural Soil: Redbrown clayey silt. Sampled at 2 ft.
	3-4	BG	Natural Soil: Redbrown clayey silt with trace quartz gravel. Sampled at 3.5 ft.
	4-6	BG	Natural Soil: Redbrown and white clayey silty sand with trace quartz gravel. Sampled at 6 ft.

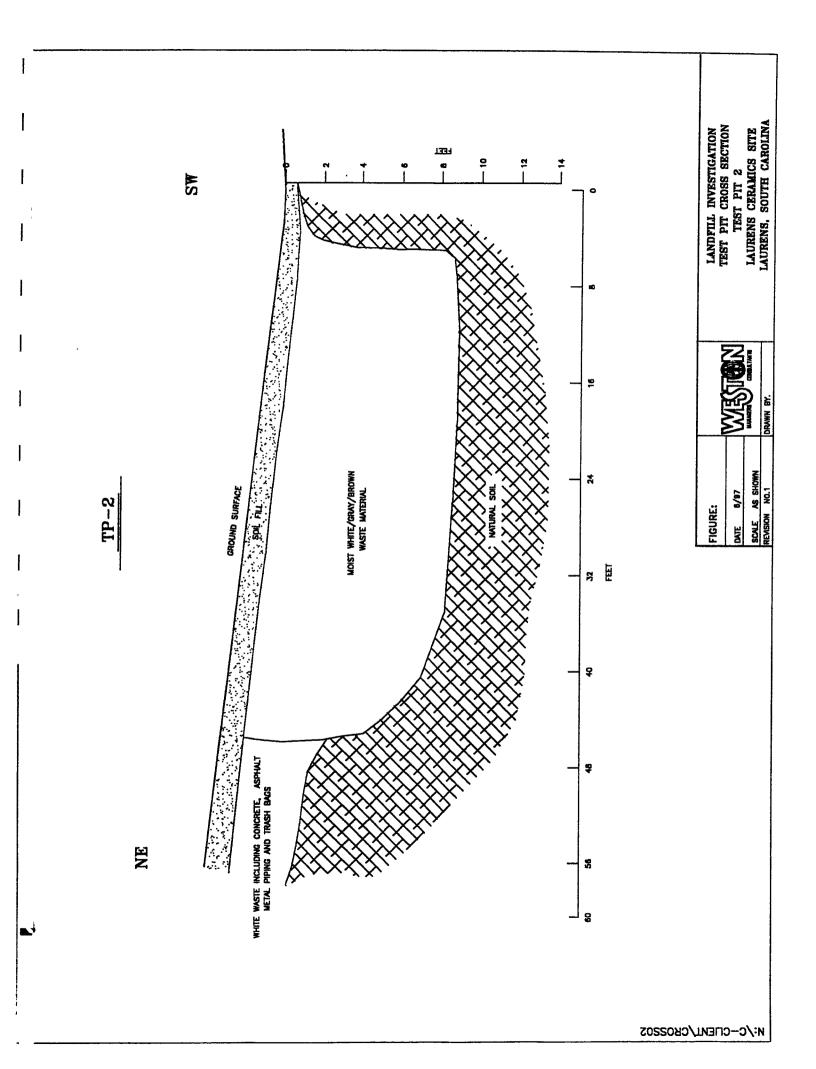
Note: BG indicates OVA measurement at background levels.

GENERALIZED SUMMARY OF TEST PIT RESULTS - SPRING 1994 LAURENS CERAMICS FACILITY LAURENS, SC

Test Pit #	Depth (Ft)	Material
TP-11	0-2	Fill: clayey silts and silty clays
	2-6	Natural soil: clayey silt
		Note: small amount of dry white fine waste material and construction debris occurred
		in the SW corner of excavation
TP-12	0-1	Fill: Clayey sand
	1-6	Natural soil: clayey silts
		Note: 4 inch seam of dry white fine ceramic material occurred in the SW
		corner of excavation
TP-13	0-6	Natural soil: silty sands underlain by fine sandy clays
TP-14	0-3	Fill: fine sandy clay
	3-6	From SW to NE White and pink dry fine waste material with construction debris
		including sheet metal, wires, fired ceramics, drum lid, and ceramic brick trending
		Into clayey silty fill with wood waste
	6-7	Natural soil with abundant root remnants trending towards the surface in NE direction.
	7-10	Natural soil sandy clays trending towards the surface in NE direction
TP-15	0-2	Fill: silty clay and clayey silt
	2-4	Fill: ceramic brick, wood waste in soil matrix (note fill became thinner as progress in
		NE direction)
	4-6	Natural soil: clayey silts
TP-16	0-0.5	Fill: clayey silts and silty clays
	0.5 - 3.5	Fill: ceramic brick, scrap metal, dry white fine ceramic material (becomes
		thinner in NE direction)
	3.5-6	Natural soil: clayey silt
TP-17	0-1	Fill: silty clay and clayey silt
	1-6	Fill: ceramic bricks, plastic, ceramic powder, occasional scrap metal
	6-7.5	Fill: scrap concrete, cinder blocks
	7.5-10	Natural soil: clayey silt
TP-18	0-0.5	Fill: clayey silt
	0.5 -8	Fill: building material scraps, scrap metal, ceramic bricks, pond sediment, and cinder blocks
	8-9	Natural soil: clayey silt (water seep at 7 ft)
TP-19	0-1	Fill: clayey silt
	1-1.5	Fill: Dry white fine ceramic material (thins out completely in NE direction)
	1.5-3	Natural soil: clayey silt
TP-20	0-1	Fill: clayey silt
**-#A	1-1.5	Fill: building material scraps, scrap metal, ceramic bricks
	1.5-4	Natural soil: clayey silt
		- ·····

Note: headspace screening of soil/waste samples indicated all measurements were at background levels.





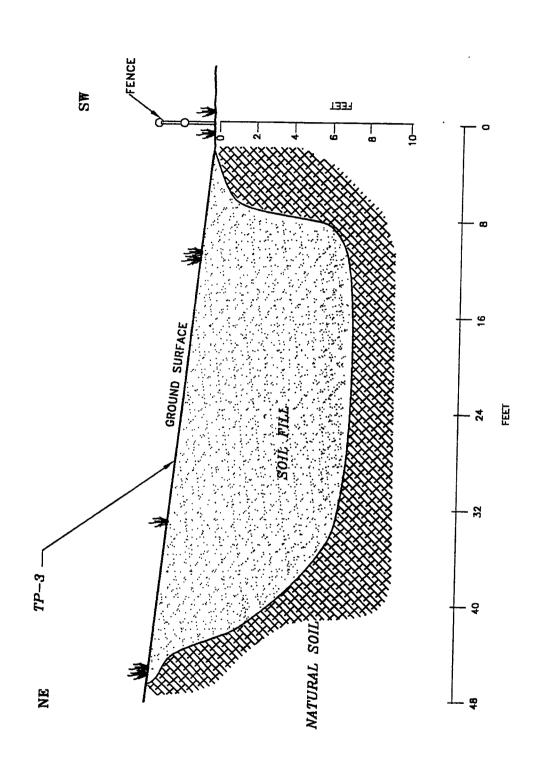


FIGURE:

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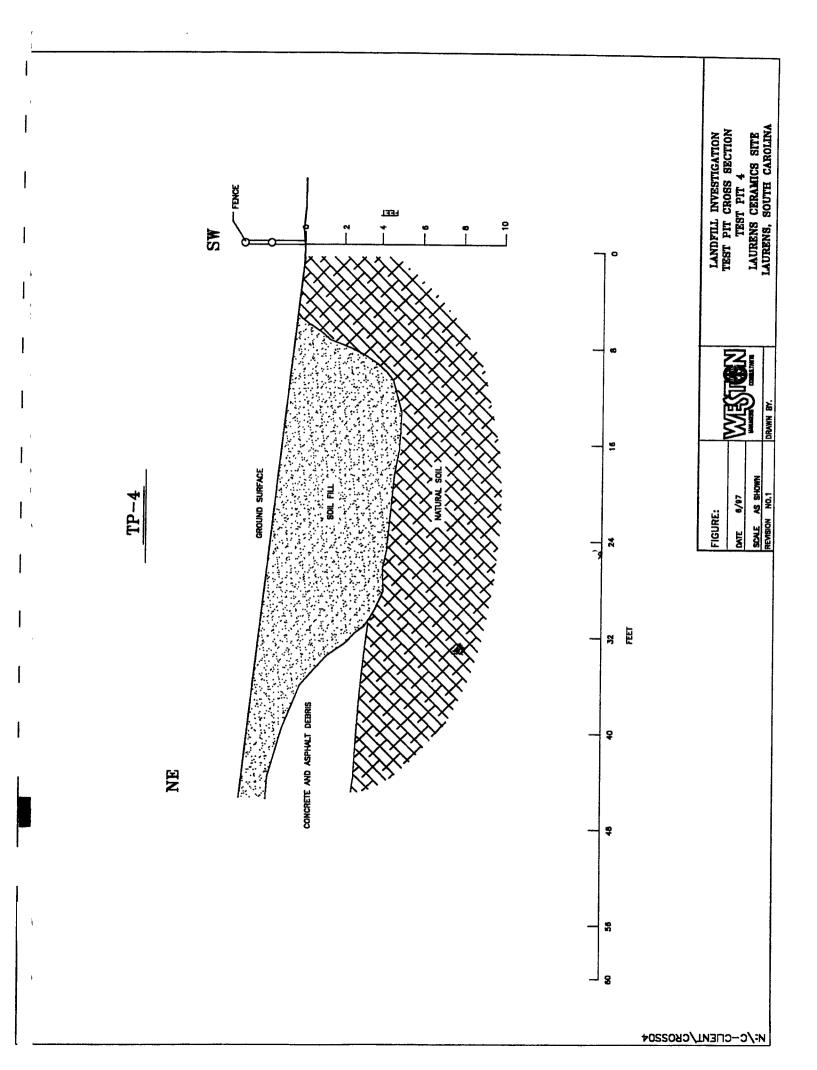
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RENSION NO.1

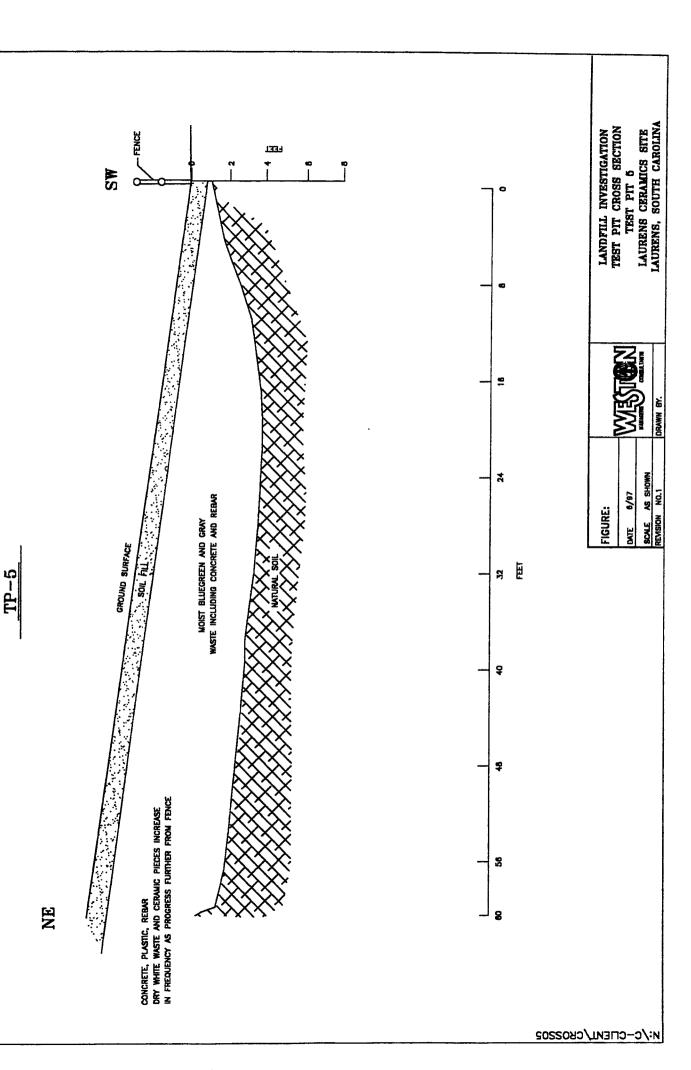
DATE OF TEST PIT 3

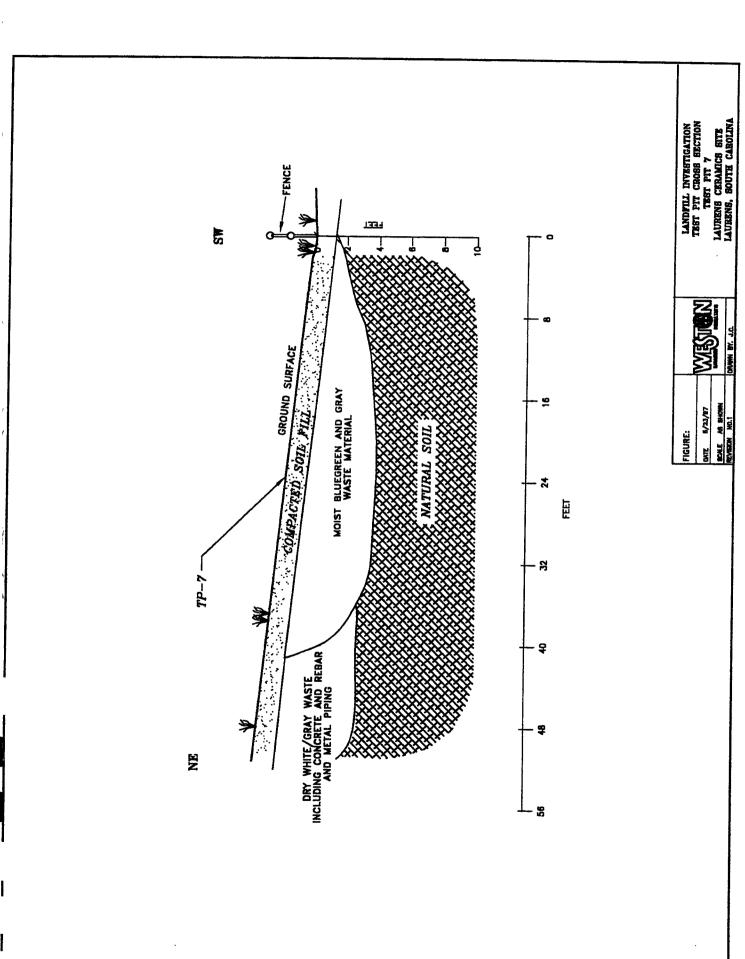
LAURENS SECTION
TEST PIT 3

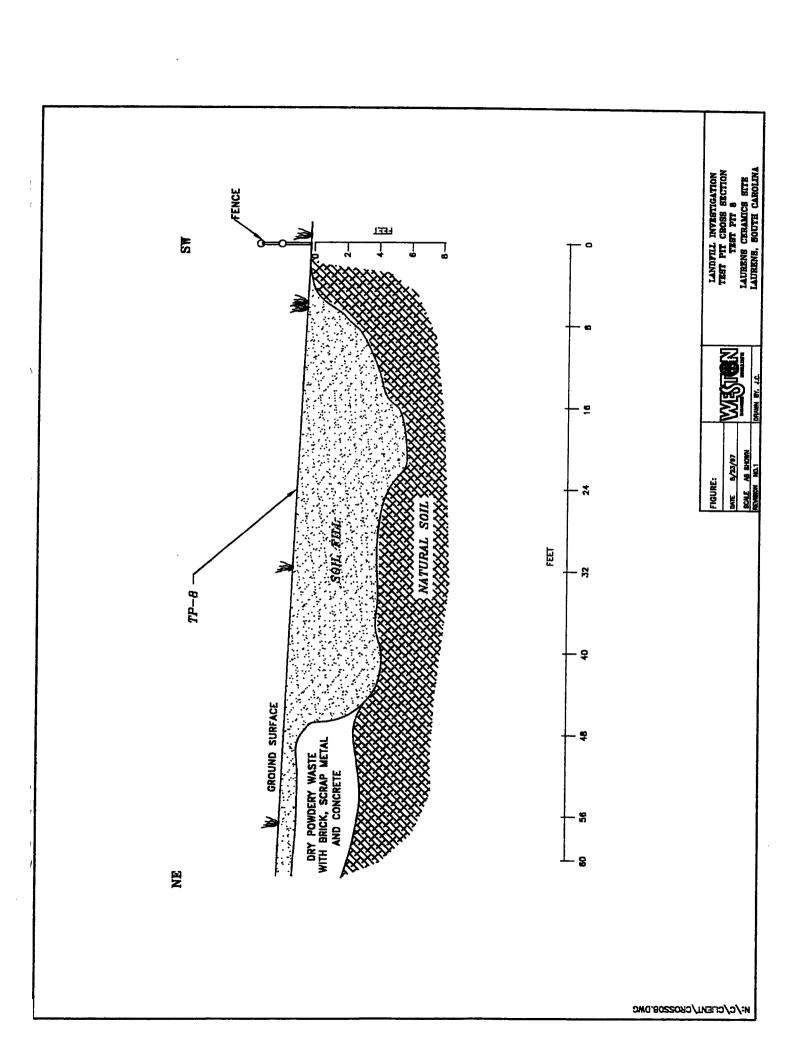
LAURENS SITE

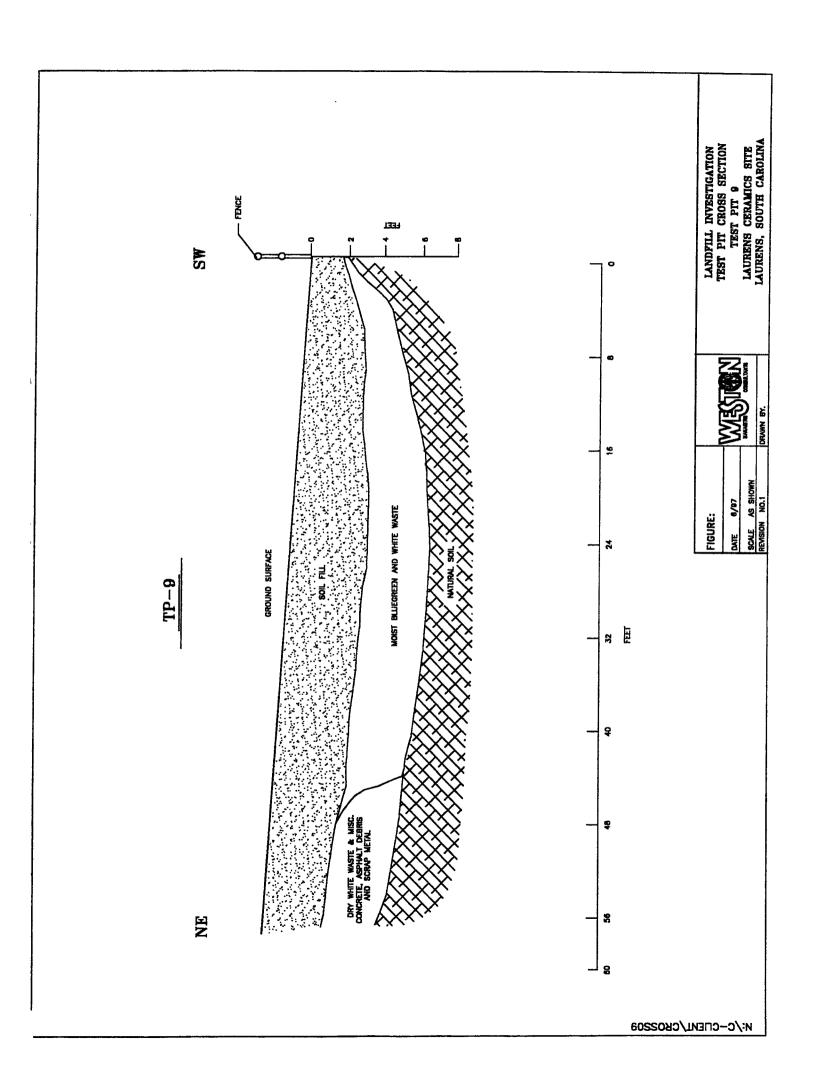
LAURENS, SOUTH CAROLINA

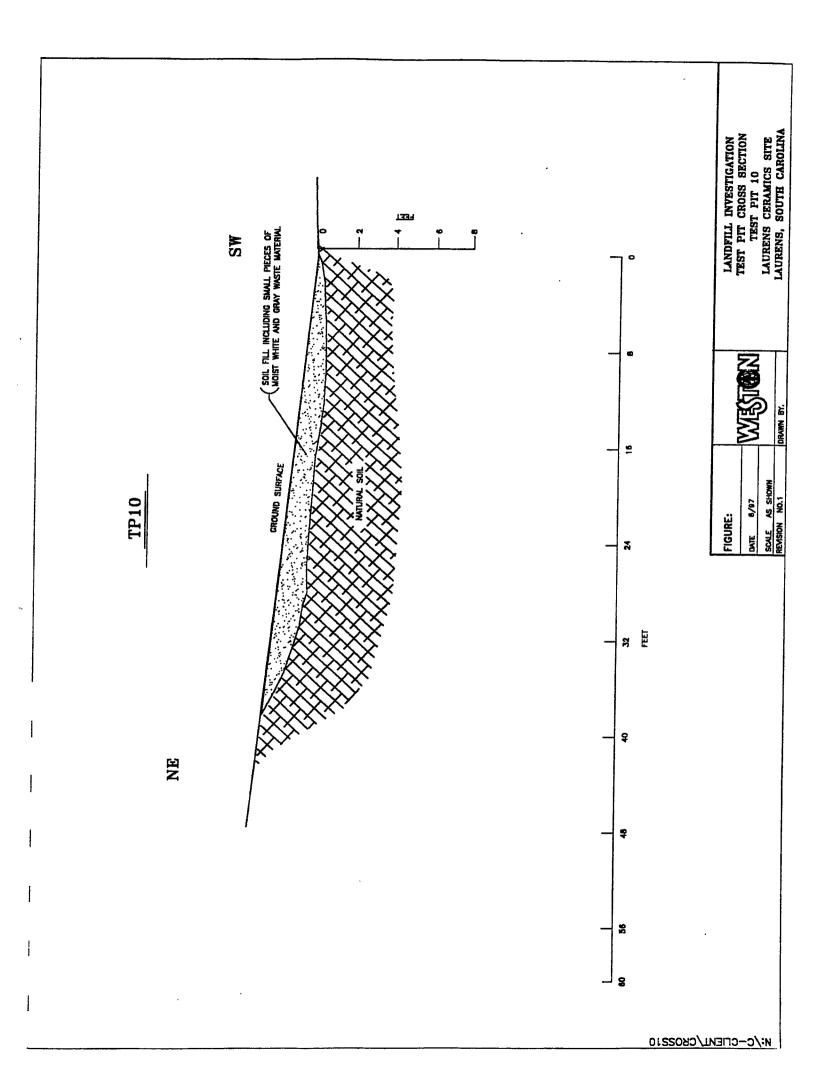


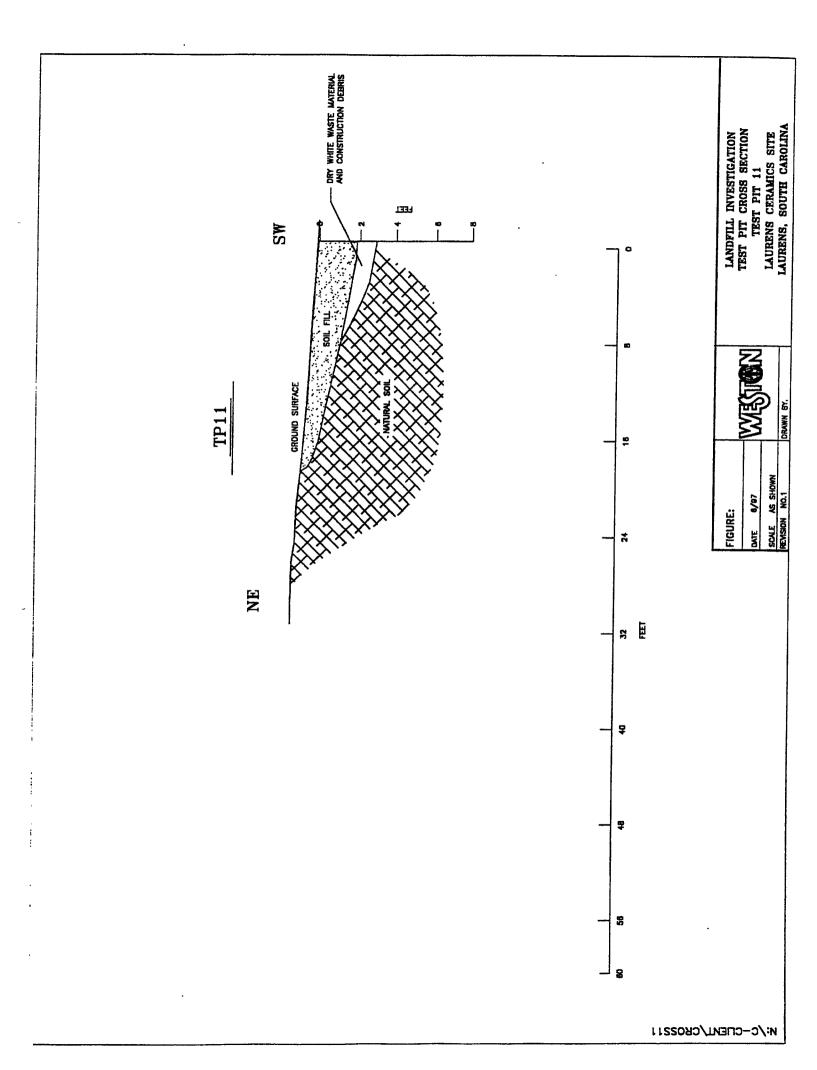


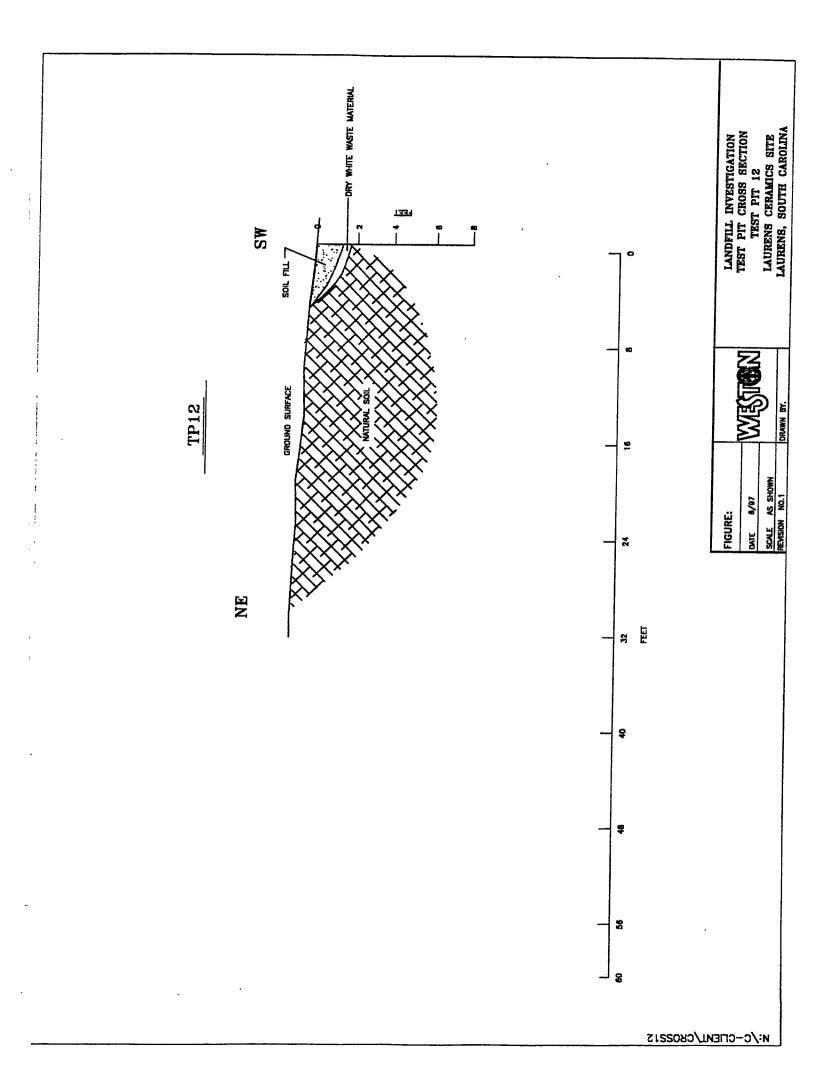


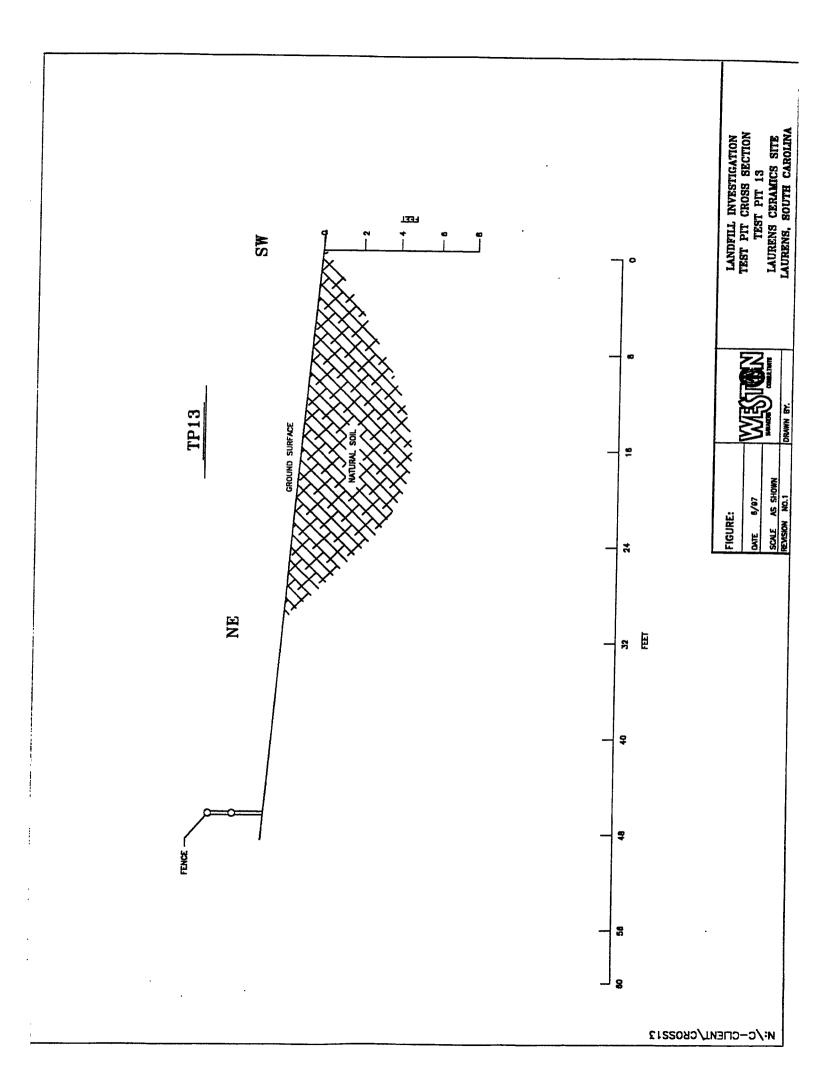


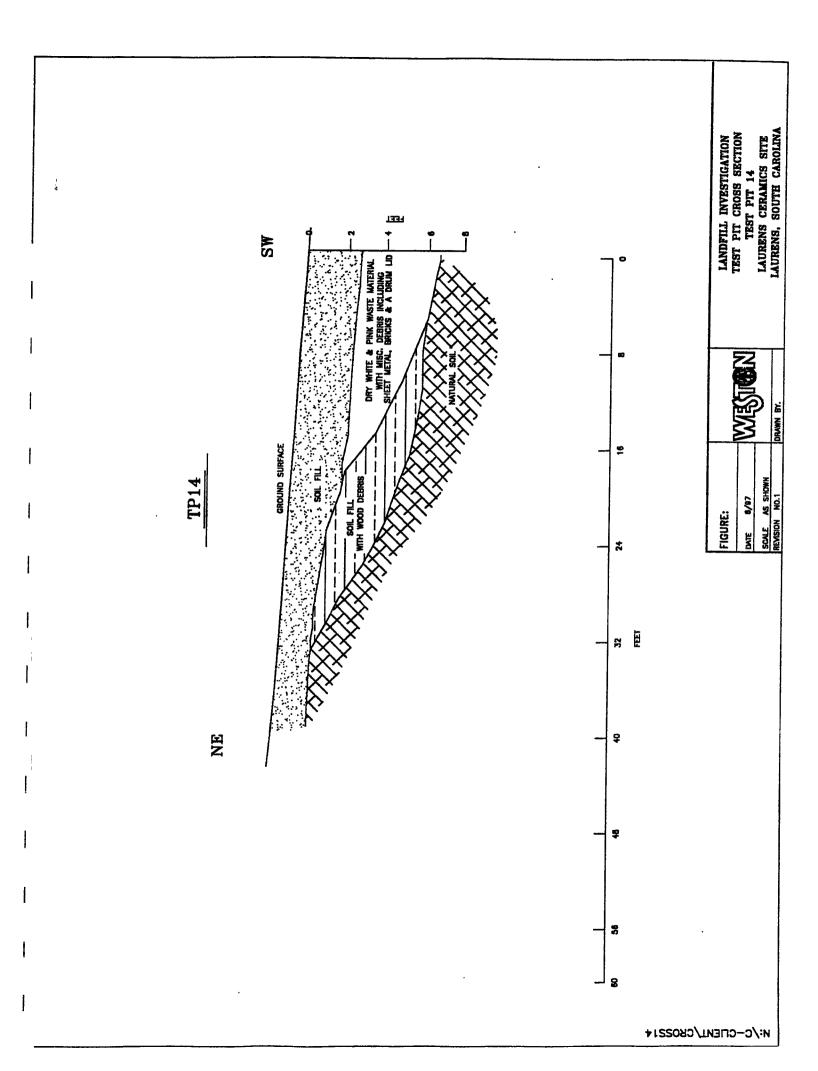


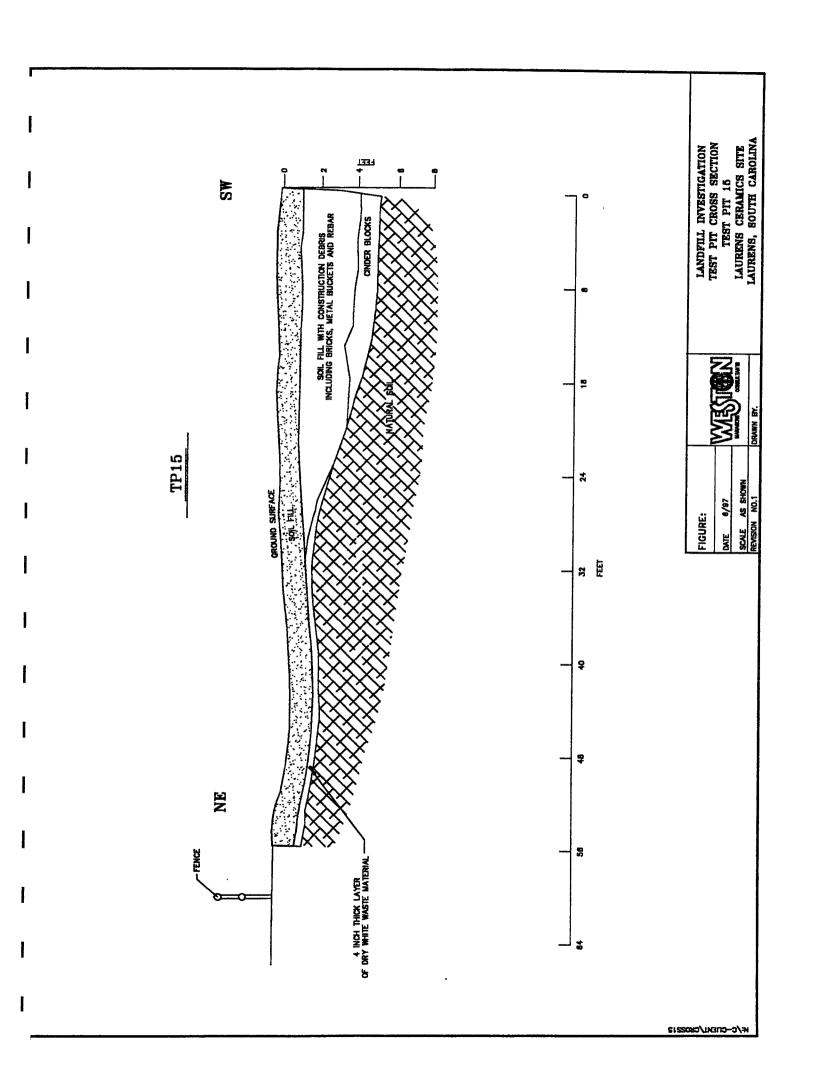


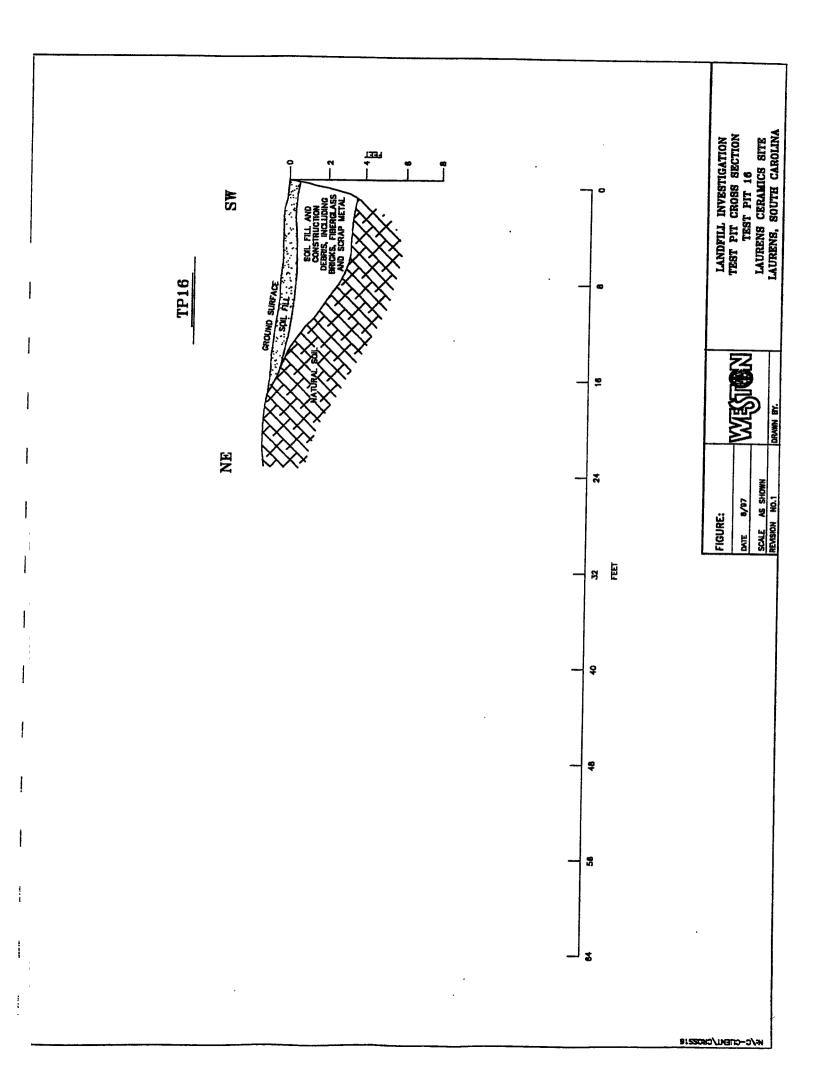


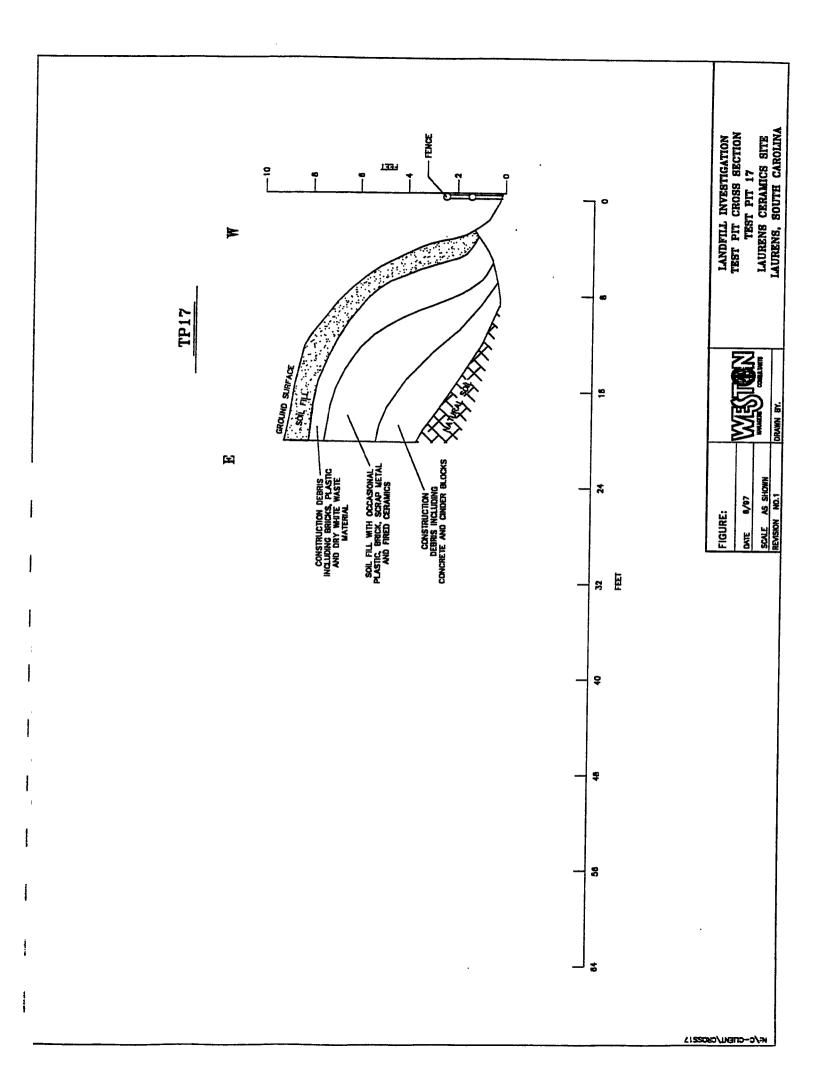


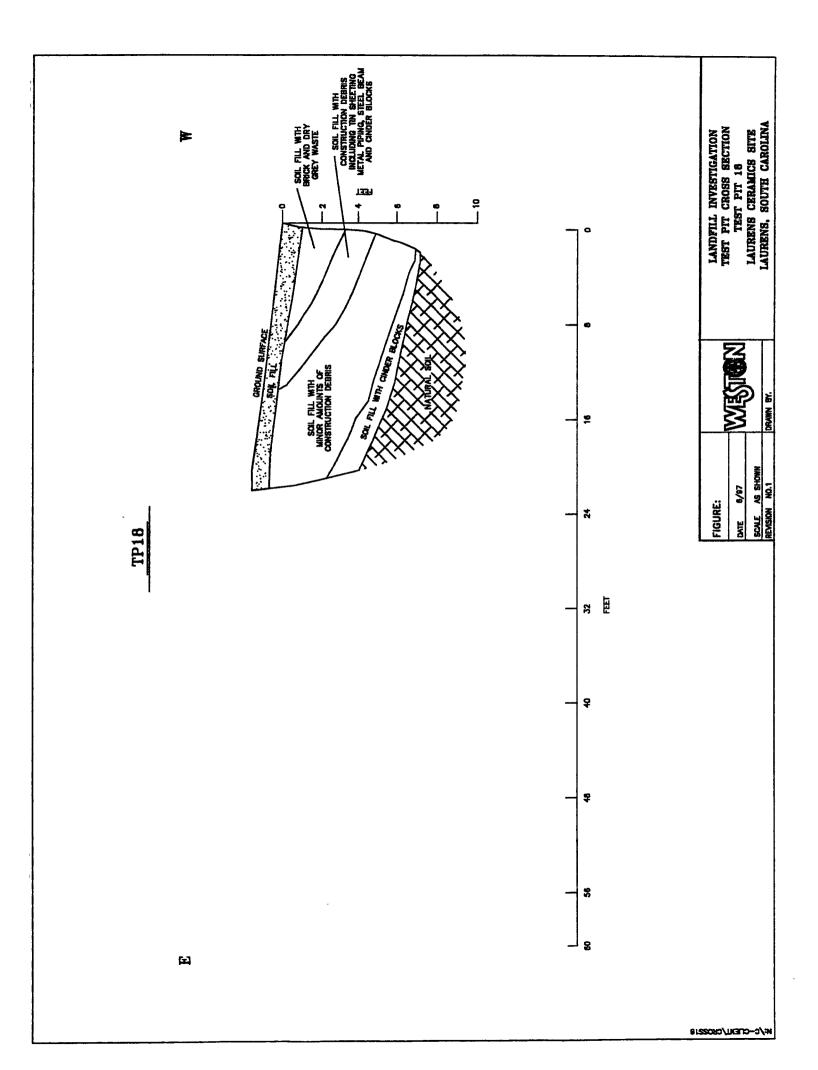


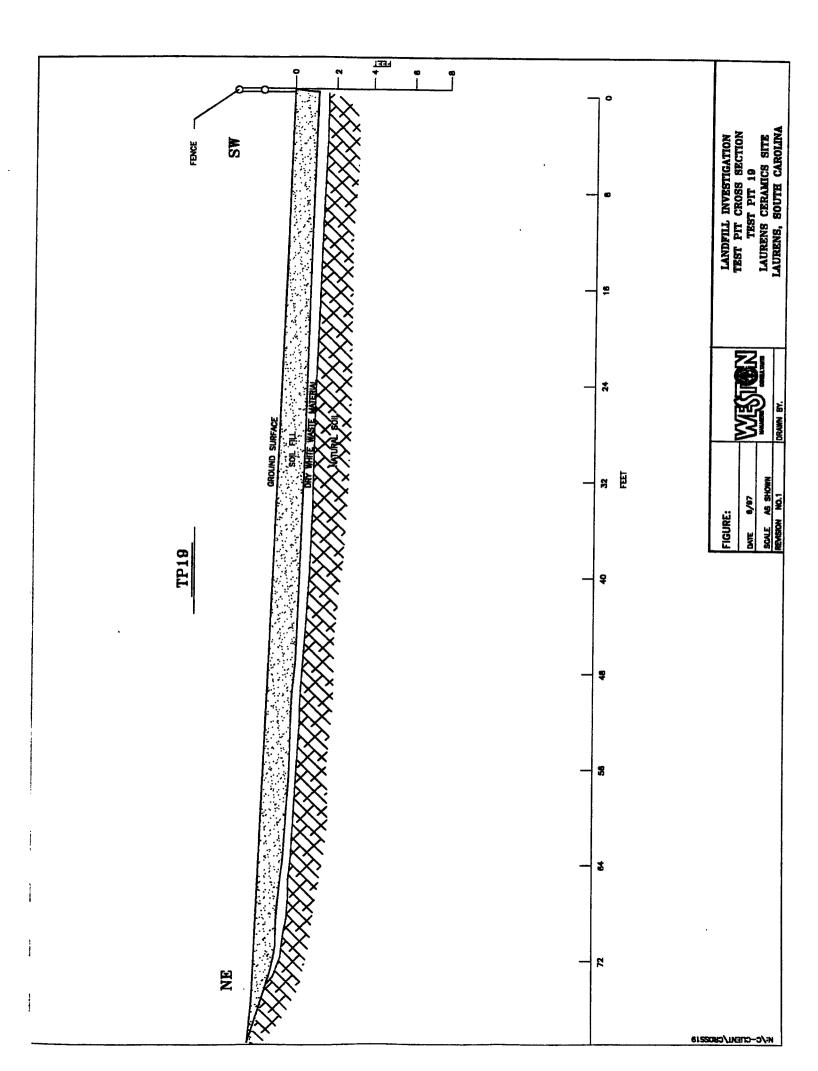


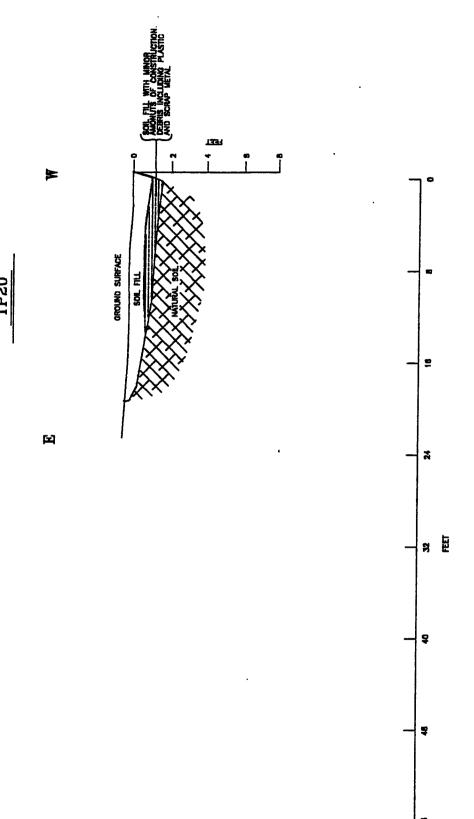












TEST PIT CROSS SECTION
TEST PIT CROSS SECTION
TEST PIT 20
LAURENS CERAMICS SITE
LAURENS, SOUTH CAROLINA

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SCULE AS SHOWN
RENSION NO.1
DEVANN BY.

FIGURE:

APPENDIX B GROUNDWATER QUALITY DATA

SUMMARY OF PURGEABLE HALOCARBONS/AROMATICS REPORTED ABOVE DETECTION LIMITS LANDFILL

MONITOR WELLS SAMPLED SEPTEMBER 1991/APRIL 1992 LAURENS CERAMICS SITE, LAURENS, SOUTH CAROLINA

PARAMETER*								Š	LOCATION							
	WW9	MW10	MW17	MW18	MW19	MW20	MW21	MW22	MW23	MW24	MW25	MW44	MW45	MW-55	MW-56	MW-57
Chloroethane	QN	QN	QN	4.0	2.3	7.8	ND	QN	QN	ND	ND	ND	ND	ND	ND	ND
Methylene Chloride	ΔN	QN	7.2	39	8.1	QN	QN	5.4	ND	ND	ND	QN	ND	ND	ND	ND
Vinyl Chloride	QN	ND	QN	5.5	11	23	αN	QN	ND	ND	ND	QN	ND	ND	ND	ND
1,1-Dichloroethene	Ω̈́	QN QN	6.5	200	7.3	12	2.4	16	QN	ND	ND	QN	ND	ND	QN	26
1,1-Dichloroethane	QN	Q	4.2	120	85	120	QN	6'9	QN	ND	1.7	QN	ND	ND	QN	34
1,2-Dichloroethane	ND	ND	UD	ND	1.1	1.6	QN	20	QN	ND	ND	QN	ND	ND	ND	5.3
Trans-1,2-Dichloroethene	QN	QΝ	5.1	2,200	420	510	6.3	4.8	ΠN	ND	15	2.8	1.3	ND	ND	220
Chloroform	QN	QN	QN	QN	αN	ND	GN	ΩN	QN	ND	QN	QN	ND	ND	ND	ND
1,1,1-Trichloroethane	QN	QN	2.2	310	4.7	13	5.4	7.0	QN	ND	ND	QN	ND	ND	ND	12
1,1,2-Trichloroethane	QN	QN	QN	QN	QN	QN	GN	QN	ΩN	ND	QN	ND	ND	ND	ND	ND
Trichloroethene	Q	GN	2.7	150	47	51	4.8	4.4	QΝ	ND	1.5	1.8	ND	ND	ND	23
Tetrachloroethene	QN	3.8	1.2	3,800	1,400	1,400	39	35	QN	ND	9.9	19	5.5	ND	1.6	130
Trichlorofluoromethane	QN	ND	QN	ND	ND	QN	QN	QN	QN	QN	QN	QN	QN	ND	QN	ND
Chlorobenzene	QN	αN	QN	3.8	QN	ND	ND	ND	ΩN	QN	QN	QN	ND	ND	QN	ND
Ethylbenzene	ND	UD	ND	ND	QN	ND	ND	ND	ND	ND	ND	QN	QN	ND	ND	ND
Benzene	ND	ND	QN	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
Toluene	ND	ND	QN	ND	ND	QN	ND	ND	ND	ND	QN	QN	ND	ND	ND	ND
Xylene (Total)	ND	QN	ND	ND	QN	QN	ND	ND	ND	QN	ND	QN	ND	ND	ND	ND

* U.S. BPA Method Nos. 8010/8020. ND indicates not detected above laboratory analytical detection limits. All values in micrograms per liter (ug/L).

MW-1 through MW-54 sampled September 5-19, 1991. MW-55 through MW-60, and MW-63 sampled April 1992.

INORGANIC DATA SUMMARY, TOTAL METALS MONITOR WELLS SAMPLED MARCH 23 THROUGH APRIL 2, 1992 LANDFILL LAURENS CERAMICS SITE LAURENS, SOUTH CAROLINA

MND ND ND									Downgradient Wells	ent Wells								Upgradient Wells	nt Wells
ND ND<	MW09 MW10 MW17 MW18	MW17		MW	18	MW19	MW20	MW21	MW22	MW23	MW24	MW25	MW44	MW45	MW55	MW56	MW57	MW40	MW41
ND	ON ON ON ON	QN		QN		QN	QN	QN	QN	ND	ND	ND	QN	ND	ON	QN	QN	QN	QN.
ND ND<	555 ND 13500 1920 (ND) (ND) (ND)	13500 (ND)		1920 (ND)		NO	QN	1690 (ND)	009	QN	QN	QN	QN	QN	602 (ND)	QN O	QN	QN	g
ND ND<	ON ON ON ON	QN		QN		ND	ND	ND	ND	ND	ND	ND	ND	QN	QN	ND	QN	Q	N N
ND ND<	QN	316		ND		ND	ND	ND	QN	ND	ND	ND	ND	ND	QN	ND	ND	QN	S
ND ND<		ON		QN		ND	ND	QN	ND	ND	ND	ND	ND	ND	QN	QN	ND	Ð	S
ND ND<	5920 ND ND ND ND	QN		ON		2580	ND	ON	5030	ND	ND	19400	ND	QN	ND	QN	8760	QN	S
ND ND ND ND ND ND ND ND	ON ON ON ON	QN		QN		ND	ND	ND	ON	QN	ON	ND	QN	QN	QN	ND	QN	Q.	Ð
ND ND<	ND ND 103 14.8 (ND)	103 (ND)		14.8		ND	QN	19.0	15.2	QN	QN	QN	QN	QN	QN	QN	QN	Q	<u>S</u>
ND 4580 840 ND ND ND 138 794 102 227 ND ND (ND) (ND) ND ND ND ND ND ND 6440 78.1 21.2 187 ND	ND	QN QN	ND			ND N	ND	ND	ND	ND	ON	ND	QN	ND	QN	ND ND	Ð	QZ	Ð
ND	ND 18100 3890	18100 3890	3890			Q	ND	4580	840	ON	QN	QN	ND	138	794	102	227	QN	130
ND	(UN)	(UN)	+	(AN)	ļ			(IN)	(GN)	,					(CN)				
6440 78.1 21.2 187 ND ND ND ND S2.8 46.1 94.0 ND (6100) (ND) (ND) (167) ND	ON ON	QN		QN	- 1	₽ E	Q	Q	ΩN	0.39	QN	QN	QN	QN	QN	QN	QZ	ND	ND
(6100) (ND) (ND) (167)		731	_	286		428	6440	78.1	21.2	187	Q	Q	S	QN	52.8	46.1	94.0	Q	ND
40.4 ND N	(148) (520)	(220)	-	(286)	,	(411)	(6100)	(QN)	(ND)	(167)					(41.0)		(81.7)		
ND ND ND ND ND ND ND ND	ND ND 68.4 ND	68.4		QN QN		QN	40.4	QN	QN	ND	QN	Ñ	ND	QN	ND	ND	ND	ND	ND
ND ND ND ND ND ND ND ND	ND ND 3.0 ND	3.0		QN		ND	ND	ND	ΩN	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
ND ND<	ON ON ON ON	QN	-	QN		ND	QN	UD	QN	ND	ΩN	ON	ND	QN	ND	ND	ND	ND	ND
ND ND<	ND ND ND ND	ND		Ω		QN	N ON	ND	QN	ND	ND	ND	ND	QN	QN	ND	ND	ND	ND
22.3 20.4 ND	ON ON ON ON	QN		QN		QN	ND	ND	QN	ND	ON	QN	ND	ND	ND	ND	ND	ND	ND
	ND ND 54.8 40.5	54.8	\dashv	40.5		QN	22.3	20.4	Q.	Q.	QN	QN	Ð	QN	QN	ND	ND	ND	ND

All values in micrograms per liter (µg/L).

ND indicates not detected above laboratory analytical detection limits.

(#) indicates the resulting concentration (µg/L) from analysis of an additional sample which was filtered through a 0.45 micron filter.

SUMMARY OF PURGEABLE HALOCARBONS/AROMATICS REPORTED ABOVE DETECTION LIMITS* MONITOR WELLS SAMPLED SPRING 1994

LAURENS CERAMICS SITE LAURENS, SC

MW17* MW18** MW19** MW20** MW22** MW22** MW22** MW22** MW22** MW22** MW22** MW23** MW41** MW35 MW35 MW35 MW36	PARAMETER (4.9/L)						Loca	Location					
ND 412 ND		MW17 ^A	MW18 ^A	MW19 ^A	MW20 ^A	MW22 ^A	MW25	MW44 ^A	MW45	MW55	MW56	MW57 ^A	MW61 ^A
ND	Chloroethane	QN	4.2	QN	QN	QN	QN	QN	ΩN	ΩN	QN	ΩN	QN
ND 15 0.544 1.6 ND ND ND ND ND ND ND N	Methylene Chloride	QN	ND	QN	QN	ND	ND	QN	QN	QN	QN	QN	QN
8.2 360 1.9 4.0 24 ND ND <td< td=""><td>Vinyl Chloride</td><td>ND</td><td>15</td><td>0.54J</td><td>1.6</td><td>ND</td><td>QN</td><td>Q</td><td>Q</td><td>QN</td><td>QN</td><td>QN</td><td>QN</td></td<>	Vinyl Chloride	ND	15	0.54J	1.6	ND	QN	Q	Q	QN	QN	QN	QN
48 200 10 22 ND	1,1-Dichloroethene	8.2	360	1.9	4.0	24	ND	Q	QN	ΩÑ	S	ND	6.9
ND ND<	1,1-Dichloroethane	48	200	10	22	QN	ND	QN	QN	Q	QN	ND	18
ND ND<	1,2-Dichloroethane	QN	QN	QN	QN	11	ND	QN	QN	Ð	Q.	2.8	QN
ND ND<	Chloromethane	Ð	Ð	QN	QN	Q	QN O	QN	QN	Ð	QN.	QN	ND
ND ND<	1,1,2,2-Tetrachloroethane	Q	ND.	QN	ND	QN	QN	Q.	Q	Q	QN	Q	ND
ND ND<	Carbon Tetrachloride	ND	QN	QN	ND	ND	ND	QN	QN	Q	Q	Ð	QN
ND 7.8 0.611 1.5 ND ND <t< td=""><td>Вготобогт</td><td>QN</td><td>ND</td><td>QN</td><td>ND</td><td>QN</td><td>ND</td><td>QN</td><td>QN</td><td>QN</td><td>Æ</td><td>Ð</td><td>QN</td></t<>	Вготобогт	QN	ND	QN	ND	QN	ND	QN	QN	QN	Æ	Ð	QN
38E 8100E 423E 1,900E 420 28E 34 ND ND ND ND	trans-1,2-Dichloroethene	ND	7.8	119:0	1.5	ND	ND	QN	ND	QN	QN	Ð	QN
ND ND<	cis-1,2-Dichloroethene	38E	8100E	423E	1,900E	420	28E	34	ND	QN	QN	320	210
11 460 21 1.1 27 1.2 0.51J S.9 ND ND 13.5 ND ND ND ND ND ND ND ND 13.5 81 6.2 13 69 ND	Chloroform	QN	ND	QN	ND	ND	ND	ΩN	QN	ND	QN	Q.	QN
ND ND<	1,1,1-Trichloroethane	21	460	21	1.1	27	1.2	0.51J	5.9	QN	QN	9.7	2.9
3.5 81 6.2 13 69 ND N	1,1,2-Trichloroethane	ND	ND	QN	ND	ND	ND	QN	ND	QN	QN	-QZ	QN
ND ND<	Trichloroethene	3.5	81	6.2	13	69	ND	9.2	ND	QN	QN	34J	08
ND ND<	Tetrachloroethene	7.5	1,000	56	092	310	7.3	130	1.6	QN	QN	69	310
ND ND<	Dichlorodifluoromethane	ΩN	ND	ND	ND	ND	ND	ND	ND	QN	QN	QN	QN
ND ND<	Trichlorofluoromethane	ND	ND	QN	ND	ND	ND	ΩN	QN	ND	Ą	Q	ND
ND ND<	Bromomethane	ND	QN	QN	ND	QN	ND	QN	QN	QN	QN	QN	QN
ND ND<	Chlorodibromomethane	ND	ND	ND	ND	ND	ND	QN	QN	QN	QN	QN	QN
ND 1.5 ND ND ND ND ND ND ND N	trans-1,3-Dichloropropene	ND	ND	QN	ND	ND	ND	ON	QN	QN	QN	Ð	QN
ND ND ND ND ND ND ND ND	Chlorobenzene	ND	1.5	ΠN	ND	ND	ND	QN	QN	QN	QN	QN	QN
N	1,2-Dichloropropane	ND	ND	QN	ND	ND	ND	0.54J	QN	QN	QN	QN	QN
N	Ethylbenzene	QN	ND	QN	ND	ON	ND	QN	ND	Ð	Ę	QN	ND
UN GN	Benzene	ND	ΩN	QN	ND	QN	ND	ΩN	QN	QN	Q	QN	QN
ON ON ON ON ON ON ON ON ON	Toluene	ND	3.8	QN	ND	ND	ND	QN	ND	QN	QN	ND	QN
	Xylene (Total)	QN	QΝ	QN	ND	QN	ND	ND	QN	ND	ND	ND	ND

ND indicates analyte was not detected at or above the analytical detection limite.

E indicates the range of calibration curve was exceeded; therefore, the results are estimated.

J indicates estimated value, analyte was detected below analytical detection limit.

Monitor wells were sampled March 22-30, 1994.

A Original analysis performed within holding time, subsequent dilutions/analyses performed past holding times.

METALS DATA SUMMARY - LANDFILL MONITOR WELLS SAMPLED SPRING 1994 LAURENS CERAMICS SITE LAURENS, SOUTH CAROLINA

	MW61	QN	QN	Q	Ð	Q	4.4	Q	QV	N ON	QN	QN	QX	Ð	2.6	0.052	3.1	QN	ND	QN	QN	QN	QZ	QN
	MW57	QN	QN	QN	ND	QN	5.5	QN	QN	ΩŽ	Q	QN	Q.	QN.	3.2	0.012	3.4	ND	QN	ON	QN	Q	QN	ND
	MW56	QN	QN	ND	QN	QN	3.7	ND	QN	QN	QN	QN	QN	ND	1.4	ND	4.1	ND	ND	ND	ND	QN	QN	ND
	MW55	QN	ND	ND	ND	ON	3.7	QN	ND	ND	QN	QN	ND	ND	1.6	ON	4.0	ND	ND	ND	ND	QN	QN	ND
	MW45	ND	ND	ND	ND	ND	4.3	QN	ND	ND	ND	ND	ND	ND	2.6	ND	3.6	ND	ND	QN	ND	ND	ND	ND
	MW44	ON	QN	ND	ND	QN	0.15	ND	ND	QN	ND	ND	ND	ND	96.0	ND	1.5	ON	ND	ND	QN	QN	ND	ND
Location	MW25	QN	ND	ON	ND	ND	19.2	ND	ND	ND	ND	ND	ND	4.2	4.4	0.012	5.7	ON	QN	QN	0.0092	ND	ND	QN
	MW22	QN	ON	QN	0.074	ND	4.6	ND	ON	ND	ND	QN	UD	ND	3.0	0.014	1.6	ND	ND	ND	ND	QN	ND	QN
	MW20	QN	ND	ND	0.13	ON	2.7	ND	0.38	ND	ND	ND	ND	QN	2.7	6.1	3.2	0.055	ND	QN	ND	ND	ND	0.031
	MW19	ND	ND	ND	0.069	ND	6.0	ND	ND	ND	ND	ND	ND	ND	4.8	0.77	3.6	Q	ON	QN	QN	QN	ND	ND
	MW18TOT	ND	0.34	ND	0.054	QN	0.20	ND	ND	ND	ND	0.41	ND	ΩN	2.0	0.042	3.2	ND	ND	ND	ND	QN	ND	0.020
:	MW18	QN	ND	ND	0.053	QN	0.13	ΩN	ND	ND	ND	ND	QN	ND	2.1	0.041	3.4	ND	QN	ND	QN	ΩN	ND	0.024
	MW17	ND	ND	ON	0.055	QN	2.8	ND	QN	ND	ND	ND	ND	ND	3.3	0.50	1.2	ND	ON	ND	QN	ND	ND	ND
	PARAMETER (mg/L)	Silver	Aluminum	Arsenic	Barium	Beryllium	Calcium	Cadmium	Cobalt	Chromium	Copper	Iron	Mercury	Potassium	Magnesium	Manganese	Sodium	Nickel	Lead	Antimony	Selenium	Thallium	Vanadium	Zinc

All samples have been filtered through 0.45 micron filter, except those labeled as "TOT". ND indicates analyte was not detected at or above the analytical detection limit. Monitor wells were sampled March 22-30, 1994. Samples were analyzed at WESTON's laboratory.

ANALYTICAL DATA SUMMARY MONITOR WELLS SAMPLED SPRING 1994 LAURENS CERAMICS SITE LAURENS, SOUTH CAROLINA LANDFILL

							Location						
PARAMETER	MW17	MW18	MW18TOT	MW19	MW20	MW22	MW25	MW44	MW45	MW55	MW56	MW57	MW61
Alkalinity	24.0	QN	NA	34.0	27.0	24.0	66.0	ND	25.0	14.0	19.0	29.0	24.0
BOD 5 Day	QN	ND	NA	ND	ND	ND	ND	ND	ND	ND	ND	QN	QN
Chloride	ND	5.3	NA	4.9	5.0	ND	ND	2.0	2.1	QN	ND	QN	Ð
Cyanide	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ΩN	ND
Fluoride	QN	0.11	NA	ND	QN	ND	0.11	QN	ND	ND	0.20	ND	QN
Hardness	20.8	ND	ND	31.0	16.7	21.1	66.2	ND	19.4	16.1	13.0	24.0	19.0
Nitrate, as N	0.21	0.20	NA	0.38	0:30	0.40	ND	0.20	0.35	08.0	0.42	0.41	S.
TKN	ND	ON	NA	1.7	ON	0.85	QN	QN	ND	ND	ND	QN	0.85
Total Organic Carbon	ON	ND	NA	2.0	2.0	ND	ON	ND	ND	ND	ND	QN	Q.
Sulfide	ND	ND	NA	ND	QN	ND	QN	ND	ND	ND	ND	ND	QN
Sulfate	5.6	ND	NA	ND	QN	ND	7.7	7.4	ND	ND	7.0	8.7	5.5
Total Dissolved Solids	2/2	30	NA	84	62	52	120	36	66	86	70	08	62
Total Organic Halogens	0.071	4.1	NA	0.42	0.32	99.0	ND	0.084	0.011	0.011	0.035	0.22	0:30
Total Suspended Solids	190	ND	NA	&	14	ND	21	30	ND	81	34	18	110
Fecal Coliform Bacteria	ND	ND	NA	ND	ND	ND	ND	QN	ND	ND	ND	QN	ND

NA indicates not analyzed.

ND indicates analyte was not detected at or above analytical detection limit.

All values in milligrams per liter (mg/L) except fecal coliform bacteria (colonies/100 ml).

Monitor wells were sampled March 22-30, 1994.

Samples were analyzed at WESTON's laboratory.

FIELD PARAMETERS MONITOR WELLS SAMPLED SPRING, 1994 LANDFILL LAURENS CERAMICS SITE LAURENS, SC

						ÓΊ	LOCATION					
FIELD PARAMETER	MW-17	MW-18	MW-19	MW-20	MW-22	MW-25	MW-44	MW-45	MW-55	MW-56	MW-57	MW-61
Dissolved Oxygen (mg/L)	2.0	4.0	3.2	3.0	3.0	0.9	7.8	4.5	5.5	0.9	5.0	5.5
Eh (mv)	212	262	324	291	192	208	254	264	239	207	178	172
Temperature (°C)	17.3	17.2	17.5	17.3	18.5	19.9	16.8	17.0	17.2	16.2	16.4	15.6
Hd	5.75	5.80	5.7	9.60	29.9	8.4	5.32	5.9	6.9	6.87	7.0	96.9
Specific Conductance (mS)	0.054	0.034	0.097	080.0	0.059	0.163	0.027	0.064	0.054	0.051	0.073	0.065
Turbidity (NTU)	20	3	NA	NA	5	1.28	19.3	12.0	155	49	17	150

PURGEABLE HALOCARBONS/AROMATICS REPORTED ABOVE DETECTION LIMITS MONITOR WELLS SAMPLED SPRING 1995 LAURENS CERAMICS SITE LAURENS, SOUTH CAROLINA

MW78 R £ 130 2 £ 270 S 2 S S 36 **MW77** Ð 2 £ 54 S g 250 S S 2 56 22 MW61 S 2.6 J Ð S Ð S £ S £ S 9.3 120 2 7 MW57 £ S £ 5 S £ 2.9 1.2 4 26 77 2 2 LOCATION MW56 0.66 J S S S S S S S 2 £ g S S 2 **MW55** S S 2 £ S N S 2 S S 2 S £ B 2 2 **MW25** 0.97 J E S S ΩŽ £ S S S S 1.2 1.2 ΩN 2 Ξ **MW20** 90, 3.6 J 7.6 皇 £ £ S S S S S g S S 2 8 MW18 4,600 150 J 120 £ £ £ 170 £ £ 13.1 8 630 QN 25 51 trans-1,2-Dichloroethene cis-1,2-Dichloroethene PARAMETER (µg/L) 1,1,1-Trichloroethane 1,1-Dichloroethene 1,1-Dichloroethane 1,2-Dichloroethane Methylene Chloride Tetrachloroethene Trichloroethene Vinyl Chloride Chloroethane Chloroform Freon 113 Toluene Methane Benzene

ND indicates analyte was not detected at or above the analytical detection limit.

J indicates estimated value, analyte was detected below analytical detection limit.

Monitor wells were sampled May 8-23, 1995.

Methane was analyzed at Southwest Research Institute. All other samples were analyzed at WESTON's laboratory.

METALS DATA SUMMARY MONITOR WELLS SAMPLED SPRING 1995 LAURENS CERAMICS SITE LAURENS, SOUTH CAROLINA

Parameter					Location	uc			
	MW18	MW20	MW25	MW55	MWS6	WMS7	MW61	<i>LL</i> MW	MW78
Aluminum ND	Q	QN	QN	QN	QN	QN	QN	QN	QN
Barium 0.063	63	0.11	ND	ND	QN	QN	QN	QN	QN
Calcium 0.22	22	1.7	15.9	3.6	3.5	5.2	3.7	10.1	8.9
Iron 0.046	946	QN	QN	QN	ND	QN	QN	0.092	0.059
Iron (Fe ²⁺) ND	D	ND	ND	ND	QN	QN	ΩN	ΩN	QN
Iron (Fe ³⁺) ND	D	ND	ND	ND	ND	QN	QN	0.070	QN
Potassium ND	D Q	ON	3.9	ND	ND	QN	QN	2.6	2.0
Magnesium 2.4	4	2.0	3.8	1.6	1.3	3.1	2.3	4.2	3.4
Manganese 0.10	10	3.3	ND	ND	ND	GN	ND	0.017	0.015
Sodium 3.7	7	2.7	5.1	3.8	3.8	3.1	2.9	4.8	4.3

All samples have been filtered through 0.45 micron filter.

ND indicates analyte was not detected at or above the analytical detection limit.

Monitor wells were sampled May 8-23, 1995.

Iron (Fe²⁺) and Iron (Fe³⁺) were analyzed at Southwest Research Institute.

All other samples were analyzed at WESTON's laboratory.

ANALYTICAL DATA SUMMARY MONITOR WELLS SAMPLED SPRING 1995 LAURENS CERAMICS SITE LAURENS, SOUTH CAROLINA

Parameter (mg/L)					Location				
	MW18	MW20	MW25	MW55	MW56	MW57	MW61	WW77	MW78
Alkalinity	QN	20.9	64.8	23.0	25.1	29.3	18.8	43.9	41.8
BOD 5 Day	Q	£	QN	QN	QN	QN	QN	QN	QN
Chloride	9.3	3.8	ND	QN	QN	ND	QN	2.1	QN
Fluoride	QN	QZ	0.10	QN	ND	ND	ND	QN	QN
Hardness	10.4	12.3	55.4	15.4	14.2	25.6	18.6	42.6	31.1
Ammonia, as Nitrogen	QN	Q.	QN	QN	ND	QN	ND	ND	QN
Nitrite, as N	QN	ΩN	QN	QN	QN	ND	QN	ΩN	QN
Nitrate, as N	0.35	0.39	ND	0.82	0.44	0.44	QN	ND	0.13
Total Organic Carbon	1.4	Ð.	QN	QN	QN	ΩN	QN	ND	QN
Sulfide	ΩÑ	QN QN	QN	QN	QN	QN	ND	ND	QN
Sulfate	5.3	ΩN	8.0	6.2	QN	7.0	8.6	10.1	6.4
Total Dissolved Solids	18	44	94	92	36	99	89	96	94
Carbon Dioxide (free)	NC	107	0.11	33	50	37	119	59	113

ND indicates analyte was not detected at or above analytical detection limit. NC indicates not calculated due to low pH (<5). Monitor wells were sampled May 8-23, 1995. Samples were analyzed at WESTON's laboratory. Carbon dioxide was calculated from analytical results provided by Weston's laboratory.

DRAFT TRITIUM DATA SUMMARY Monitor Wells Sampled Spring 1995 Landfill

Laurens Ceramics Site Laurens, South Carolina

Location	Tritium (T.U.)	Detection Limit (T.U.)
MW18	20.0	0.7
MW20	18.4	0.7
MW25	0.9	0.8
MW55	11.3	0.7
MW56	11.7	0.7
MW57	8.0	0.7
MW61	19.6	0.6
MW77	15.8	0.6
MW78	19.2	0.8

Surface waters were sampled May 8-23, 1995.
All samples were analyzed at University of Georgia's Isotope Laboratory.

APPENDIX C SURFACE WATER QUALITY DATA

Summary of Volatile Organic Compounds in Surface Water Samples Laurens Ceramics Site Laurens, South Carolina Landfill

Sample Location:	WSW07	WSW07	WSW07	WSW07	WSW07	WSW07	WSW07	WSW07	WSW07	WSW07	WSW07	WSW07
Date:	9/20/91	10/11/91	8/19/92	11/17/92	2/17/93	5/7/93	8/10/93	10/27/93	3/31/94	6/29/94	9/16/94	12/1/94
1,1,1-Trichloroethane	QN	QN	QN	QN	0.3.J	ND	QN	ND	GN	QN	ND	ND
1,1,2,2-Tetrachloroethane	QN	ON	QN	QN	QN	QN	ON	ND	N	ND	QN	QN
1,1,2-Trichloroethane	QN	QN	QN	QN	QN	QN	ND	ND	ND	ND	QN	QN
1,1-Dichloroethane	ND	QN	QN	QN	QN	QN	ND	ND	ND	CIN	QN	QN
1,1-Dichloroethene	ON	ON	QN	ON	QN	QN	ON	ON	ND	QN	QN	QN
1,2-Dichlorobenzene	ND	QN	N N	ON	QN	ON	ON	ND	N	ND	S S	ON
1,2-Dichloroethane	ND	ON	ND	ND	QN	ND	ND	ND	QN	QN	QN	ON O
1,2-Dichloroethene (total)	NA	NA	NA	QN	NA	NA	NA	NA	NA	NA	νγ	NA
1,2-Dichloropropane	ND	QN	QN.	ON	QN	ON	ND	ND	ON	ND	QN	ON
1,3-Dichlorobenzene	ND	QN	QN	ON	QN	ON	ON	QN	QN	NO	QN	QN
1,4-Dichlorobenzene	ND	QN	QN	ND	QN	ND	N	N N	QN	QN	QN.	QN
2-Chloroethylvinylether	QN	QN	NA	ON	QN	ON	ND	ND	QN	NO	ND	QN
Benzene	ND	QN	QN	ND	ND	ND	ND	ND	QN	ON	QN	QN
Bromodichloromethane	ND	QN	QN	ON	ND	QN	ND	ΩN	QN	ON	QN O	ON
Bromoform	ON	QN	QN	ND	ND	ND	ND	QN	QN	QN O	Q.	CN CN
Bromomethane	QN	QN	QN	ON	ON	QN	ND	NO	ND	QN	QN	QN
Carbon Tetrachloride	ND	QN	ND	QN	QN	QN	ND	ND	ON	QN	QN	QN
Chlorobenzene	ND	QN	QN	ND	ON	ND	ND	ND	QN	QN	ND	QN
Chlorodibromomethane	ND	QN	QN	QN	QN	N	ND	QN	QN	ND	ΩN	QN
Chloroethane	ND	QN	QN	ND	ND	ND	ND	ND	QN	QN	QN	QN
Chloroform	ND	QN	ON.	ND	ND	ND	N	ON	QN	QN	QN	QN
Chloromethane	ND	QN	QN	ON.	ND	ND	N	QN	ON.	QN	QN	QN
cis-1,2-Dichloroethene	NA	NA	NA	NA	NA	NA	NA	NA	NA	ΝΑ	νγ	QN
cis-1,3-Dichloropropene	ND	QN	QN	N	ND	ND	ND	ND	QN	ΩN	QN	ND
Dichlorodifluoromethane	ND	ND	Vγ	ND	ND	ND	ND	ND	QN	ON	QN O	ND
Ethylbenzene	ON	ΩN	QN.	ND	ND	ND	QN	ND	Q	ΩN	QN O	QN
Freon 113	Vγ	ΝΑ	NA	NA	NA	NA	NA	NA	NA	ΩN	Ϋ́Α	QN
Methane	NA	NA	NA	NA	NA	Ϋ́	NA	NA	NA	NA	NA	NA
Methylene Chloride	2	QN	QN	ND	3.7 B	ND	ND	CN	ON ON	ON ON	0.78 JB	QN
MIBK (4-methyl-2-pentanone)	ΝΑ	ΝΑ	ΝΑ	ΝĄ	NA	NA	NA	NA	NA	NA	ΝΑ	ON O
Tetrachloroethene	ND	ΩN	QN ON	ND	7	ND	QN	QN	QN	.	QN	QN
Toluene	ND	QN	QN	ND	ND	NO	ON	ON	QN	Q	QN	QN
trans-1,2-Dichloroethene	N	ND	QN	ND	0.44.3	ND	ND	N N	QN	QN	QN	QN
trans-1,3-Dichloropropene	ND	ΩN	QN ON	NO	ND	ND	QN	ON	ON	QN	QN	QN
Trichloroethene	ND	ND	ON.	N	ND	ON	QN	QN O	QN	QN	QN.	QN
Trichlorofluoromethane	ND	QN	Q.	N	ND	QN	QN	QN	QQ	ON	QN	ND
Vinyl Chloride	ND	NO	QN	N	ND	ON N	QN	ON	Q.	QN	QN O	QN
Xylene	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
All results reported in microprams per liter												

All results reported in micrograms per liter.

ND = Not detected.

NA = Not analyzed.

J = estimated value, analyte was detected below analytical detection limit.

B = Compound was also detected in blank.

Second of the properties of the properti

Summary of Volatile Organic Compounds in Surface Water Samples Landfill

Laurens Ceramics Site Laurens, South Carolina

Sample Location:	WSW07	WSW07	80MSM	80MSM	WSW08	WSW08	WSW08	WSW08	WSW08	WSW08	WSW08	WSW08
Date:	2/1/95	5/24/95	9/20/91	10/11/91	2/8/92	5/18/92	8/19/92	11/11/92	2/17/93	5/7/93	8/10/93	10/27/93
1,1,1-Trichloroethane	ND	QN	QN	QN	ON	QN	ND	QN	0.2.3	ND	QN ,	QN
1,1,2,2-Tetrachloroethane	ND	QN	QN	QN	QN	QN	QN	ON	ΩÑ	ND	NO	QN
1,1,2-Trichloroethane	QN	QN.	QN	ND	ON	ND	QN	QN	0.22.3	QN	ON	QN
1,1-Dichloroethane	ND	QN	QN	ND	ON	QN	QN	QN	Ð.	QN	QN	QN
1,1-Dichloroethene	QN	QN	QN	QN	ON	QN	ND	ON	QN	ND	QN	QN
1,2-Dichlorobenzene	QN	QN	QN	QN	QN	ON	QN	QN	QX	QN	QN	QN
1,2-Dichloroethane	ND	QN	QN	QN	QN	QN	QN	QN	QN.	QN	ON	QN
1,2-Dichloroethene (total)	NA	NA	NA	NA	NA	NA	NA	QN	NA	NA	Ϋ́N	NA
1,2-Dichloropropane	QN	QN	ND	QN	ON	ON	ND	QN	QN.	QN	NO	QN
1,3-Dichlorobenzene	QN	QN	ND	QN	QN	QN	QN	QN	QN ON	QN	ND	ND
1,4-Dichlorobenzene	QN	QN	QN	ND	ON	ON	ND	QN	QN	ON	QN	QN
2-Chloroethylvinylether	QN	QN	QN	QN	ON	ON	NA	QN	QN.	QN	QN	ON
Benzene	QN	QN	ND	QN	ON	ND	ON	QN	QN	QN	ND	ON
Bromodichloromethane	QN	QN	ND	QN	ND	QN	ND	QN	QN	ON	ND	ON
Bromoform	QN	QN	QN	QN	ON	ON	ND	QN	QN.	QN	ND	QN
Bromomethane	QN	QN	ND	QN	QN	ON	QN	QN	QN	QN	ND	ND
Carbon Tetrachloride	ND	QN	ND	QN	QN	ON	NO	QN	QN	ND	ND	ON
Chlorobenzene	ND	QN	QN	QN	ON	QN	QN	QN	QN.	QN	QN	QN
Chlorodibromomethane	QN	QN	QN	ND	ON	ND	ND	ND	QN	QN	QN	QN
Chloroethane	QN	QN	QN	QN	ON	QN	QN	QN	QN	QN	ON	QN
Chloroform	QN	QN ON	QN	QN	QN	QN	ND	QN	QN	QN	QN	QN
Chloromethane	QN	ON.	ND	QN	QN	QN	ND	ND	QN	ON	ON	QN
cis-1,2-Dichloroethene	QN	QN	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA
cis-1,3-Dichloropropene	NO	QN	ND	QN	QN	ON	QN	QN	QN	QN	QN	ON
Dichlorodifluoromethane	QN	QN	QN Q	QN	ON	ND	NA	QN	QN	QN	QN	ND
Ethylbenzene	ND	ON ON	NO ON	QN	QN	ND	QN	QN	QN	QN	ND	ON
Freon 113	OZ OZ	Q Q	۸N	VΝ	NA	VV	ΝΑ	NA	NA	ΝΑ	VΛ	NA
Methane	ΥN		Ϋ́	NA	NA	NA	ΝΑ	NA	NA	NA	NA	NA
Methylene Chloride	QN ON	0.56.1	Q Q	4.6	ON	ΩN	QN	QN	3,68	QN	QN	ON
MIBK (4-methyl-2-pentanone)	QX	NA VA	ΑN	ΝΑ	NA	NA	NA	NA	NA	NA	NA	NA
Tetrachloroethene	ON ON	Q	77	QN	ON	QN	QN	ΩN	QN	QN	QN	QN
Toluene	QX	0.77.3	Ð	QN.	N O N	4	QN	QN	QN	QN	QN	ON
trans-1,2-Dichloroethene	QN	Q.	1.3	N N	ON	QN	QN	NA	QN	QN	QN	QN
trans-1,3-Dichloropropene	QN ON	Ω	ND	QN	ON	ON	QN	QN	QN	QN	QN	QN
Trichloroethene	QN	Q.	QN	QN	QN	QN	QN	QN	QN	QN	QN	ON
Trichlorofluoromethane	ND	QN	ND	QN	QN	ND	ND	QN	QN	QN	QN	ND
Vinyl Chloride	QN	Q.	Q.	QN.	ON	ON	QN	QN	QN	QN	QN	QN
Xylene	ND	ND	ND	ND	ND	ND	ND	ON	ND	ND	QN	ND
All results renorted in micrograms ner liter												

All results reported in micrograms per liter.

ND = Not detected.

NA = Not analyzed.

J = estimated value, analyte was detected below analytical detection limit.

B = Compound was also detected in blank.

Compound was also detected in blank.

Summary of Volatile Organic Compounds in Surface Water Samples Laurens Ceramics Site Laurens, South Carolina Landfill

Date: 33194 62994 91694 121194 1.1Trichlorocthane ND ND ND ND 1.1Trichlorocthane ND ND ND ND 1.1Dichlorocthane ND ND ND ND 1.2Dichlorocthane ND ND ND ND 1.3Dichlorocthane ND ND ND ND 2.Chlorocthane ND ND ND ND Bromonethane ND ND ND ND	7	10/1	27	2/8/92 C C C C C C C C C C C C C C C C C C C	8/19/92 CM CM CM CM CM CM CM CM CM CM CM CM CM	ON ON ON ON ON ON	2/17/93
chloroethane ND ND ND chloroethane ND ND ND chloroethane ND ND ND oroethane ND ND ND sthane ND ND ND oroethane ND ND ND oroethane ND ND ND oroethane ND ND ND oro				<u> </u>	999999	9999	-
etrachlorocthane				<u> </u>	22222	222	
orocethane ND ND ND orocethane ND ND ND orochene ND ND ND orochazene ND ND ND orochonzene ND ND ND orobenzene ND ND ND orbenzene ND ND ND methylvinylether ND ND ND morbenzene ND ND ND oromomethane ND ND ND oromomethane ND ND ND oromomethane ND ND ND					9999	22	
orocethane ND ND ND orochene ND ND ND orochenzene ND ND ND orochane ND ND ND orochenzene ND ND ND orochenzene ND ND ND orochenzene ND ND ND orochenzene ND ND ND sthylvinylether ND ND ND m ND ND ND chlarene ND ND ND m ND ND ND methorochene ND ND ND chlorochene ND ND ND				<u> </u>	222	Q !	
orocethene				<u> </u>	ON ON		
orobenzene ND ND ND oroethane ND NA NA NA oroethane ND ND ND ND orobenzene ND ND ND ND schloromethane ND ND ND ND schloromethane ND ND ND ND mane ND ND ND ND scene NA <t< td=""><td></td><td>•</td><td></td><td><u> </u></td><td>QN</td><td>2</td><td></td></t<>		•		<u> </u>	QN	2	
orocethane ND ND ND orochane (otal) NA NA NA orochene (total) ND ND ND orobenzene ND ND ND orobenzene ND ND ND orobenzene ND ND ND orobenzene ND ND ND orochonzene ND ND ND orbiloromethane ND ND ND oromomethane ND ND ND oromomethane ND ND ND oromomethane ND ND ND oromomethane ND ND ND oridlorochene ND ND ND oridlorochene ND ND ND oridlorochene ND NA NA orothoride ND ND ND oromomethane ND ND ND oromomethane ND NA				£ £ £ £ £ £ £ £	!	ND	
orocethene (total) NA NA NA orobenzene ND ND ND ethylvinylether ND ND ND m ND ND ND				<u> </u>	QN	QN	
oroporopane ND ND ND orobenzene ND ND ND orobenzene ND ND ND orobenzene ND ND ND sthylvinylether ND ND ND m ND ND ND shloromethane ND ND ND sthane ND ND ND strane ND ND ND rane ND ND				<u> </u>	NA	NO	
orobenzene				<u> </u>	QN	QN	
orobenzene ND ND ND ethylvinylether ND ND ND m ND ND ND m ND ND ND etrachloride ND ND ND nzene ND ND ND rane ND ND ND richorocethene ND ND NA rane ND ND ND richorocethene ND ND ND rane ND ND ND rane ND ND ND rane ND ND ND				9999	ND	ND	
ethylvinylether ND ND ND m ND ND ND m ND ND ND etrachloride ND ND ND sthane ND ND ND racene ND ND ND racene ND ND ND rane ND ND ND <				999	QN	QN	_
ND				<u> </u>	NA	QN	_
ND				CIN	QN	ND	
MD M				- בעני בעני	QN	ON	
MD M				QN	QN	ND	j.
MD M				QN	QN	QN	_
MD M				QN	QN	QN	
MD ND				QN	QN	QN	
MD ND				QN	QN	QN	
MD ND				ND	QN	ND	
MA NO				QN !	QN !	QN !	
MAN NO				QX ;	Q ;	OZ :	
MD ND				VA	NA	NA	
				Q !	Q ;	Q ;	
				QN !	VY.	Q ;	
				Q ;	ON ;	Q ;	
	ON S	V V	AN AN	V X	Y X	V Z	
				£ £	C Z	S S	0.86.IB
		Y V		N AN	NA VA	N AN	
ON O				ND	QN	QN	
ON O	QN		UN UN	QN	QN	ND	
ON O		QN		QN	QN	NA	
ON O				QN	QN	QN	
CIN				QN	QN	QN	
incurance in the last of the l				QN	QN	QN	
ON ON				QN —	QN	N Q	
Xylene ND ND ND ND ND	ND	ND	ND ND	ND	ND	ND	
All results reported in micrograms per liter. ND = Not detected. NA = Not analyzed. Le stimated value, analyte was detected below analytical detection limit						^	
B = Compound was also detected in blank. Concentraction detected and reported by							

Page 3 of 7

Summary of Volatile Organic Compounds in Surface Water Samples I,andfill

Laurens, South Carolina Laurens Ceramics Site

Date:	5/7/93	WSW09	WSW09	3/31/94	WSW09	WSW09	WSW09	WSW09	WSW09	WSW09	WSW09	WSW13
1 1 1-Trichloroethane	CIN	CIN	CIN	NIN	NID	AID AID	MD 4	ND AID	01.44175	11/20/23 JID	0/14/20	16/11/01
1,1,1-11101101101111110	QV.	ON!	ON !	2	Q.	QN	QN	ON.	N N	ND	NO.	ON
1,1,2,2-Tetrachloroethane	ON	QN ON	QN	ON	NO	ND	QN	ΩN	QN	ND	QN	QN
1,1,2-Trichlorocthane	S S	QN	QN.	QX	ND	QN	QN	QN	NO	ND	QN	ON.
1,1-Dichloroethane	QN	ND	QN	ON	ND	ND	QN	QN	QN	QN	QN	QN
1,1-Dichloroethene	CIN	ND	QN	QN	QN	ND	QN	QN	QN	QN	QN	S
1,2-Dichlorobenzene	QN	ND	QN	QX	N	ND	QN	QN	QN	QN	QN	S
1,2-Dichloroethane	ON	ND	ON	ON	ND	ND	QN	QN	QN	QN	GN	C N
1,2-Dichloroethene (total)	NA	NA	NA	NA	NA	NA	ΝΑ	NA	NA	NA	Ϋ́N	NA
1,2-Dichloropropane	QN	ND	QN	QN	QN	QN	QN	QN	QX	GN	G X	S
1,3-Dichlorobenzene	QN	ND	ND	QX	QX	GN	G N	C X	S	G Z	G N	S
1,4-Dichlorobenzene	QN	ND	QN	QX	S	CN	CX	S	S	S	Ş	S
2-Chloroethylvinylether	Q.	C X	S	S	S	CN	CN	C N	S		2 2	
Benzene	Ę	C N	S	S	C Z	C N	C N		2 2	<u> </u>	2 5	<u> </u>
Bromodichloromethane	Ş	CN				2 2	S S	g g	2 5	Q (2	2 5	2 5
Bromoform	2 5	2 5	2 5	2 5	2 2	QN CN			2 2		2	2 5
Bromomethane	2	2 5	2 5	2 5	2 2	Q Z	Q X	Q Z	2 5	2	2	2
Carbon Tetrachloride	2 5	2 2	2 5	2 5	3 5	ON C	ON S	Q E	S S	Q K	Q E	2 5
Chieftent Charles Inc	2	QV S	ON S	Q ;	Q į	ON :	ON !	Q ;	Q !	ON!	QN !	ON!
Chlorobenzene	€ ;	Q ;	Q ;	Q.	QN !	QN	QN	QN	QN	QN	QN	NO
Chlorodibromomethane	CN ;	CIN !	QN	Q.	QN	QN	QN	QN	NO	ND	QN	NO
Chloroethane	Q !	QN !	ON !	ON O	Q.	QN	QN	QN	ON ON	NO NO	QN O	N
Chlorotorm	ON !	QN	QN	ON	ON O	ND	ND	QN	QN	QN	QN	N
Chloromethane	QV.	ON ON	2	ON ON	NO	QN	ON	QN	ND	GN	QN	NO
cis-1,2-Dichloroethene	NA	NA	NA	ΝΑ	NA	NA	ΩN	ΩN	ND	ND	QN	NA
cis-1,3-Dichloropropene	QN	ΩN	QN	ON.	QN	ND	QN	QN	N	ND	ND	ND
Dichlorodifluoromethane	QN	Q	ON ON	QN	QN	ND	QN	QN	QN	ND	QN	ON
Ethylbenzene	QN	N N	ON.	ON.	QN	QN	QN	QN	N	ND	N N	N ON
Freon 113	NA	NA	NA	NA	ON	NA	ON	QN	N	ND	NO	NA
Methane	NA	NA	NA	NA	NA	NA	NA	NA	ND	NA	NA	NA
Methylene Chloride	QN	QN	QN	Q.	N	1.1 B	QN	QN	N	1.1 B	ND	ND
MIBK (4-methyl-2-pentanone)	NA	NA	NA	NA	NA	NA	ON	QN	NA	NA	NA	NA
Tetrachloroethene	QN ON	ON	ON	Q.	ND	ND	ON	QN	N	ND	QN	ND
Toluene	QN	ON	ON	QN	ND	ND	QN	ND	N	ON	QN	ND
trans-1,2-Dichloroethene	QN	ON ON	QN	QN	ND	ND	QN	QN	ND	ND	QN	QN
trans-1,3-Dichloropropene	QN	QN	QN	QN	ND	ND	QN	QN	ND	ND	QN	ND
Trichloroethene	QN	N O N	QN	QN	ND	ND	QN	QN	NO	ND	QN	ND
Trichlorofluoromethane	QN O	QN	QN	QN ON	QN	ND	QN	QX	N	CIN	QN	ND
Vinyl Chloride	QN	QN ON	QN ON	Q.	QN	QN	QN	QN	N N	ND	QN	ND
Xylene	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
All results reported in micrograms per liter. NID = Not detected. NA = Not analyzed. J = estimated value, analyte was detected												
below analytical detection limit. B = Compound was also detected in blank.												
Concentraction detected and reported by												
laboratory.												

Summary of Volatile Organic Compounds in Surface Water Samples Landfill

Laurens Ceramics Site Laurens, South Carolina

Sample I contion:	WCW13	WSW13	WCW13	WCW13	WSW13	WSW13	WSW13	WSW13	WSW13	WSW13	WSW13	WSW13
Date:	8/19/92	11/17/92	2/17/93	5/7/93	8/10/93	10/27/93	3/31/94	6/29/94	9/16/94	12/1/94	2/1/95	5/24/95
1,1,1-Trichloroethane	QN	ON	QN	QN	QN	ON	1.9	QN	QN	QN	ND	QN
1,1,2,2-Tetrachloroethane	ND	ON	QN	QN	QN	ON	QN	QN	QN	QN	QN	QN
1,1,2-Trichloroethane	QN	QN	QN	N N	QN	QN	Q.	QN ON	Q.	ON.	ON.	QN
1,1-Dichloroethane	N Q	QN	ON	NO	ND	QN	ΩN	ON O	QN	ON	ON.	QN
1,1-Dichloroethene	ND	QN	QN	QN.	QN	QN	3.1	QN	QN	QN	QN.	ON
1,2-Dichlorobenzene	ON.	N N	N O	N Q	QN.	QN	Ð	Q.	QN	ON.	Q.	QN
1,2-Dichloroethane	QN	QN	QN	QN	QN	ON	QN	QN.	QN	QN	Q.	ON
1,2-Dichloroethene (total)	NA	QN	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA
1,2-Dichloropropane	ON	QN	ON	QN	QN	ND	QN.	ON.	QN	QN	QN	QN
1,3-Dichlorobenzene	ND	NB	QN	ND	QN	ND	QN	QN	Q.	Q	QN O	QN
1,4-Dichlorobenzene	ND	QN	ON	ND	Q.	ND	QN	QN	ON.	QN	QN O	QN
2-Chloroethylvinylether	NA	QN	QN	QN	Q	N N	Q.	2	2	Q !	Q !	Q !
Benzene	QN	QN	ND	ND	Q	ND	QN	Q Z	Q N	Q	ON N	QN
Bromodichloromethane	QN	QN	ON	ΩN	ON	R	QN	Q	ON N	ON ON	2	ON.
Bromoform	QN	ΩN	NO	ND	N Q	ND	QN	QN	QN	Q	Q	QN
Bromomethane	ΩN	ΩN	ND	QN	QN	ND	QN	QN	O.N.	QN	QN	QN
Carbon Tetrachloride	ON	QN	ND	ND	QN	1.5.B	QN	QN	S S	QN	ON.	ON
Chlorobenzene	ON	QN	ON	ND	QN	N N	QN	QN	QN	QN	ON	QN
Chlorodibromomethane	QN	QN	ON	ND	QN	ND	QN.	QN	QN	QN	QN	QN
Chloroethane	QN	QN	ND	ND	QN	N	ND	ND	QN	QN	QN	ND
Chloroform	N	QN	QN	ND	ON	N	ND	ON	ΩN	QN	ON.	QN
Chloromethane	QN	QN	NO	ND	QN	N	QN	QN	ΩN	QN	QN	NO
cis-1,2-Dichloroethene	NA	NA	NA	NA	NA	NA	ΝΑ	NA	NA	QN	Q.	QN
cis-1,3-Dichloropropene	ON	ND	ND	ND	ND	ND	ND	QN	ΩN	ON	QN	QN
Dichlorodifluoromethane	NA	QN	ND	ND	ND	ND	QN	QN	ΩN	QX	QN.	ND
Ethylbenzene	ΩN	QN	ND	ND	ND	ND	QN	QN	ΩN	QN	QN	ND
Freon 113	NA	NA	NA	NA	NA	NA	ΝΑ	QN	NA	ΩN	QN O	QN
Methane	NA	NA	NA	NA	NA	NA	NA	NA	Ϋ́	NA	Ϋ́	ON.
Methylene Chloride	QN	ND	ND	QN	QN	N	QN	ΩN	1.1 B	QN	QN	QN
MIBK (4-methyl-2-pentanone)	NA	NA	NA	NA	NA	NA	NA	ΝΑ	NA	Q	ON ON	NA
Tetrachloroethene	QN	QN	ND	QN	1.9	ON	ND	N Q	QN	ΩN	ON	ON
Toluene	QN	ND	ND	QN	ND	ON	ON	ND	ΩN	ΩN	ON ON	ND
trans-1,2-Dichloroethene	ΩN	NA	ND	QN	CIN	ON	ND	ND	QN	ΩN	ON	ON.
trans-1,3-Dichloropropene	ΩN	QN	NO	QN	QN	QN	QN O	ND	QN	QN	Q	QN
Trichloroethene	ΩN	QN	ND	QN	ND	ΩN	QN	QN	QN	QN	ON O	QN
Trichlorofluoromethane	ND	QN	ND	CIN	QN	ON CO	QN	ND	ON	QN	QN	ON.
Vinyl Chloride	QN	NO	N	QN	QN	QN	QN	N	QN O	Q	Q.	QN
Xylene	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND
All results reported in microprams per liter												

All results reported in micrograms per liter.

ND = Not detected.

NA = Not analyzed.

J = estimated value, analyte was detected below analytical detection limit.

B = Compound was also detected in blank.

SSC Oncentraction detected and reported by laboratory.

Summary of Volatile Organic Compounds in Surface Water Samples Laurens Ceramics Site Laurens, South Carolina Landfill

thane e local)	5 8/14/96	1011111						•			
	CIN	10/11/91	8/19/92	11/17/92	2/17/93	5/7/93	8/10/93	10/27/93	3/31/94	6/29/94	9/16/94
	2	QN	QN	ON	0.32.3	QN	QN	QN	QN	QN	ND
otal)	QN	N O	N O	QN	QN	QN	QN	QN	QN	QN	ΩN
otal)	QN	QN	QN ON	QN	QN	QN	QN	QN	ND	QN	QN
otal)	QN	QN	ND	QN	QN	QN	QN	QN	QN	CIN	CN.
otal)	QN	QN O	QN	<u>N</u>	QN	QN	QN	QN	QN	QN	QN
(total)	QN	ΩN	ND	ON O	QN	QN	QN.	QN	ND	ND	ΩN
lotal)	QN	QN ON	ND	QN	QN	QN	QN	QN	QN	QN	QN
	NA	NA	NA	QN	NA	NA	NA	NA	NA	ΝΑ	NA
	QN	ND	QN	QN	ND	ND	ND	ND	QN	QN	QN
1,3-Dichlorobenzene	QN	QN	QN	QN	QN	QN	QN	QN	QN	QN	ND
1,4-Dichlorobenzene ND	QN	ND	N N	QN	QN	QN	QN	QN	QN	QN	QN
ethylvinylether	ND	QN	QN	QN	QN	ON	QN	QN	QN	ND	ND
	ON	QN	ND	QN	QN	QN	QN	ND	QN	QN	QN
Bromodichloromethane	QN	QN	QN	QN	QN	QN	QN	QN	QN	QN	QN
	QN	QN ON	QN	NO	ND	QN	QN	QX	QN	ON ON	ND
Bromomethane	QN	QN	QN	QN	QN	QN	QN	ON	QN	ND	ND
loride	QN	QN	QN	QN	QN	QN	QN	QN	QN	QN	QN
Chlorobenzene	QN	QN	QN	QN	QN	ΩN	QN	QN QN	QX	QN	QN
omethane	QN	QN	QN	QN	QN	QN	QN	QN	QN	QN	QN
9	QN	QN	ND	<u>N</u>	ND	ON	QN	QN	QN	QN	N
	QN	QN	QN	Q.	QN	CIN	ON	QN	QN	QN	QN
	Q	Q.	QN	ΩN	QN	QN	QN	QN	QX	QN	ND
	QN	NA	NA	WA	NA	NA	NA	NA	NA	NA	NA
	QN	QN	QN	Q.	QN	QN	N Q	QN	QN.	QN	QN
oromethane	ON	QN	NA	QN	QN	ON	QN	QN	QN	ON	QN
ene	QN	QN	QN	QN	QN	ND	ND	ON	QX	ON	N N
3	QN	NA	NA	NA	NA	NA	NA	NA	NA	QN	NA
	NA	NA	ΝΑ	NA	NA	NA	NA	NA	AN	NA	NA
	QN.	QN	ON.	Q.	0.83.JB	ON	NO NO	QN	QN	Q.	1.1 B
?-pentanone)	QN	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA
oroethene	QN	QN ON	QN	ΩN	QN	ND	ND	QN	QN	QN	ΩN
	QN	QN	QN	QN	ON	ND	ND	QN	QN	ND	ND
	QN	ND	QN	NA	QN	ND	ND	QN	QN	QN	ON.
ropropene	ND	QN ON	QN	QN	ΩN	ON	ND	QN	QN	QN	ND
Trichloroethene	QN	QN	QN	QN	QN	-	ND	QN	QN	QN	QN
methane	QN	QN	QN	QN.	ON	ON	ND	QN	QN	QN	QN
hloride	QN	QN	Q	QN	QN	QN	ND	QN	QN	QN	QN
Xylene	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND

All results reported in micrograms per liter.

ND = Not detected.

NA = Not analyzed.

J = estimated value, analyte was detected below analytical detection limit.

B = Compound was also detected in blank.

SSConcentraction detected and reported by laboratory.

Landfill

Laurens Ceramics Site Laurens, South Carolina

Date: 121/194 21/195 5/24/95 5/195 5/11/28/95 11/28/95 11/28/95 11/28/95 11/28/95 11/28/95 11/28/95 11/28/95 11/28/95 11/28/95 11/28/95 11/28/95 11/28/95 11/28/95 11/28/16/10/96 11/28/16/10/96 11/28/16/16/16/16/16/16/16/16/16/16/16/16/16/	Sample Location:	WSW14	WSW14	WSW14	WSW16	WSW16	WSW16
ane ND ND ND ND ND ND ND ND ND N	Date:	12/1/94	2/1/95	5/24/95	5/1/95	11/28/95	8/14/96
ane ND ND ND ND ND ND ND ND ND N	1,1,1-Trichloroethane	QN	QN	ΩN	1.3	91	0.53
C	1,1,2,2-Tetrachloroethane	QN	ND	QN	N Q	QN	ΩN
ND	1,1,2-Trichloroethane	QN	N	QN	ON.	QN •	ND
N	1,1-Dichloroethane	QN.	N	ON	11	=	4,6
N	1,1-Dichloroethene	ND	N	ON	ON.	7.7	QN
e e la	1,2-Dichlorobenzene	QN	N N	ND	N ON	QN	QN
1	1,2-Dichloroethane	ND	N	ND	QN	QN	QN
N	1,2-Dichloroethene (total)	NA	NA	NA	NA	NA	NA
C	1,2-Dichloropropane	ND	N N	QN	ON	QN	QN
C	1,3-Dichlorobenzene	QN	ND ND	ON	ON	QN	ON
NO	1,4-Dichlorobenzene	QN	N	ON	QN	CIN	QN
ND	2-Chloroethylvinylether	QN	N O	ON	QN	ND	QN
ND	Benzene	ON	QN	QN	QN	QN	QN
ND	Bromodichloromethane	QN	ON	ON	QN	QN	ND
Colored Colo	Bromoform	QN	QN	ON	QN	QN	QN
ND	Bromomethane	QN	QN	QN	QN	QN	QN
ND	Carbon Tetrachloride	QN	ND	QN	QN	QN	QN
ND	Chlorobenzene	QN	QN	QN	QN	QN	ND
ND	Chlorodibromomethane	QN	N	ΩN	QN	QN	ON
ND	Chloroethane	QN	NO ON	QN	QN	QN	QN
ND	Chloroform	QN	N	QN	QN	QN	QN
ND	Chloromethane	QN	QN	QN	QN	CN	QN
040 040 040 040 040 040 040 040 040 040	cis-1,2-Dichloroethene	QN	QN	NO	96	25	E
010 010 010 010 010 010 010 010 010 010	cis-1,3-Dichloropropene	QN	QN	QN	Q	Ð	ND
010 010 010 010 010 010 010 010 010 010	Dichlorodifluoromethane	QN	QN	QN	QN	QN	ON
010	Ethylbenzene	QN	N	ON	QN	QN	ON
NA NA (966)	Freon 113	QN	QN	ΩN	ON	QN	ON
One) ND ND ND ND ND ND ND ND ND N	Methane	NA	NA	70	NA	Ϋ́	ON
016) ND ND ND ND ND ND ND ND ND N	Methylene Chloride	QN	QN	0.54.3	ON	QN	ON
## GN	MIBK (4-methyl-2-pentanone)	QN	QN	ΥN	NA	NA	ON
CN C	Tetrachloroethene	QN	QN	QN	Z,	87	13
ON O	Toluene	QN	ON	7,1	QN	QN.	QN
ON O	trans-1,2-Dichloroethene	QN	ON	QN	ON	QN O	ON
ND	trans-1,3-Dichloropropene	QN	QN	QN	QN	Ð	ON
MD ND ND ND ND ND ND 0,684 ND ND ND ND	Trichloroethene	QN	QN	QN	જુ	8.3	2,4
ON O	Trichlorofluoromethane	QN	ON	QN	Ð	Q	ON
ON ON ON	Vinyl Chloride	2 5	<u>R</u>	Q !	1890	0.76.3	0.78
	Xylene	QN	QN	ON	ON	ON	ND

All results reported in micrograms per liter.

ND = Not detected.

NA = Not analyzed.

J = estimated value, analyte was detected below analytical detection limit.

B = Compound was also detected in blank.

Compound was also detected in blank.

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METALS DATA SUMMARY MAY 24, 1994 SURFACE WATER SAMPLING EVENT LANDFILL LAURENS CERAMICS SITE LAURENS, SOUTH CAROLINA

Parameter (mg/L)		Loc	Location	
	WSW07	60MSM	WSW13	WSW14
Aluminum	ND	0.62	0.56	0.41
Barium	0.052	QN	ND	ΩN
Calcium	2.8	2.1	3.4	0.50
Iron	1.2	1.1	1.4	1.1
Iron (Fe ²⁺)	0.185	0.734	0.721	0.654
Iron (Fe ³⁺)	0.945	0.446	0.959	969'0
Potassium	ND	QN	QN	QN
Magnesium	1.6	1.1	1.9	0.63
Manganese	0.061	0.016	0.052	0:030
Sodium	2.0	2.6	3.0	1.0

ND indicates analyte was not detected at or above the analytical detection limit. Iron (Fe^{2+}) and Iron (Fe^{3+}) were analyzed at Southwest Research Institute. All other samples were analyzed at Weston's laboratory.

ANALYTICAL DATA SUMMARY MAY 24, 1995 SURFACE WATER SAMPLING EVENT LAURENS CERAMICS SITE LAURENS, SOUTH CAROLINA

Parameter (mg/L)		Z	Location	
	MSW07	WSW09	WSW13	WSW14
Alkalinity	QN	10.3	26.8	QN
BOD 5 Day	QN	ND	QN	QN
Chloride	2.6	2.5	2.7	2.3
Fluoride	QN	QN	0.10	0.10
Hardness	13.6	QN	16.4	ND
Ammonia, as Nitrogen	QN	ND	0.81	ND
Nitrite, as N	QN	QN	QN	ND
Nitrate, as N	QN	0.18	0.17	0.11
Total Organic Carbon	1.3	2.6	2.9	1.6
Sulfide	QN	QN	ND	ND
Sulfate	9.7	QN	ND	QN
Total Dissolved Solids	09	70	06	43
Carbon Dioxide (free)	37	21	38	NC

ND indicates analyte was not detected at or above analytical detection limit.

NC indicates not calculated due to low pH (<5).

Samples were analyzed at WESTON's laboratory.

Carbon dioxide was calculated from analytical results provided by Weston's laboratory.

DRAFT TRITIUM DATA SUMMARY

Surface Water Spring 1995 Landfill

Laurens Ceramics Site Laurens, South Carolina

Location	Tritium (T.U.)	Detection Limit (T.U.)
WSW07	16.9	0.4
WSW09	26.8	0.4
WSW13	17.2	0.4
WSW14	26.0	0.7

Surface waters were sampled May 8-23, 1995. All samples were analyzed at University of Georgia's Isotope Laboratory.